

A Tour of My Research on laser-based molecular spectroscopy

許艷珠

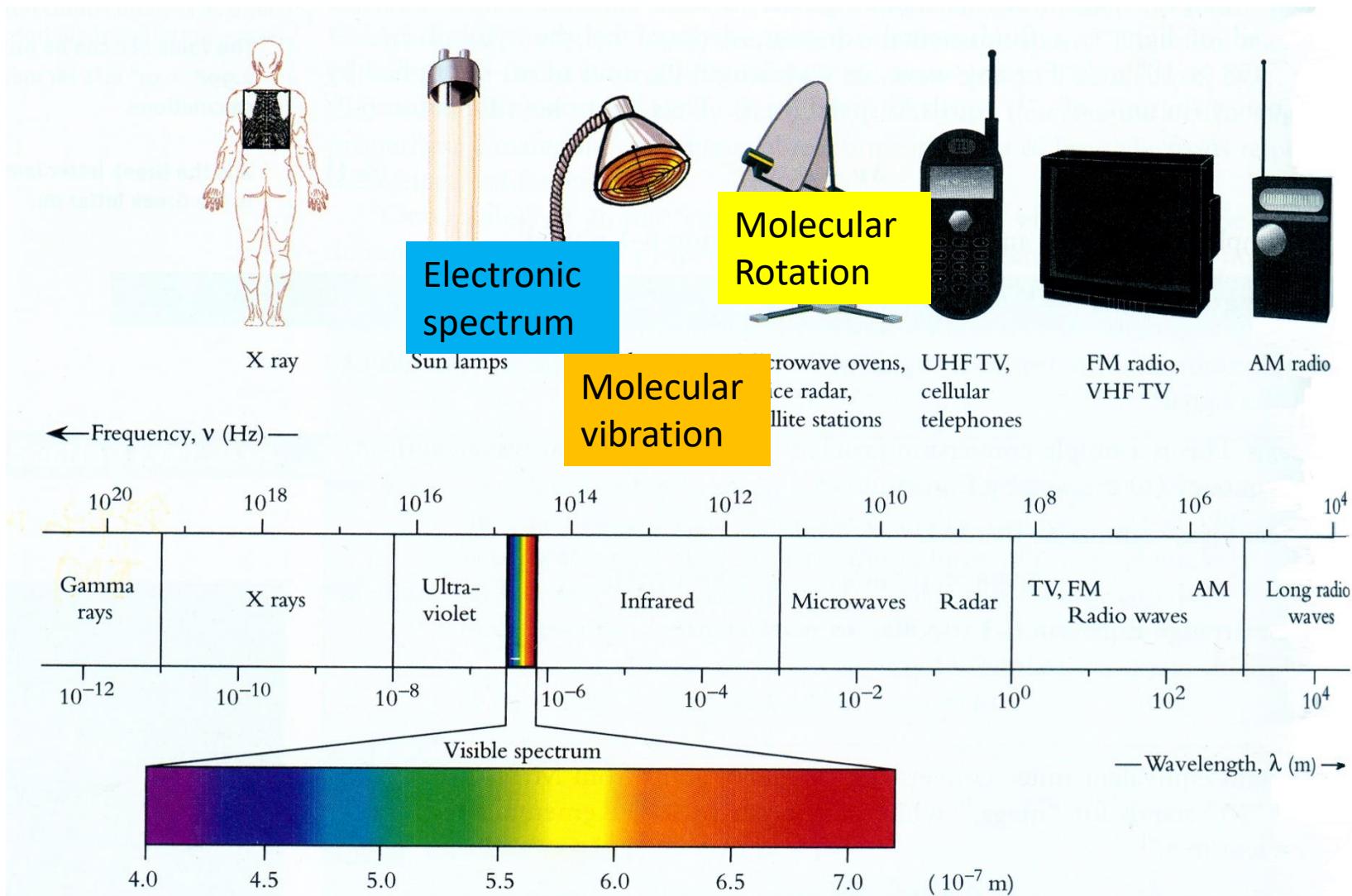
Yen-Chu Hsu

中央大學物理系

專案研究員

Lab:305室

The Electromagnetic spectrum



The Born-Oppenheimer Approximation

The full molecular Hamiltonian is

$$\hat{\mathcal{H}} = \underbrace{\frac{1}{2m_e} \sum_e p_e^2}_{\text{Electron KE}} + \underbrace{\frac{1}{2} \sum_n \frac{P_n^2}{M_n}}_{\text{Nuclear KE}} + \underbrace{V(q_e, Q_n)}_{\text{Electrostatic interactions between the charged particles}}$$

The Schrödinger equation is

$$\hat{\mathcal{H}}\Psi_{evr}(q_e, Q_n) = E\Psi_{evr}(q_e, Q_n)$$

In the Born-Oppenheimer approximation we assume that electrons, being lighter and faster than nuclei, adapt instantly to the configuration of the nuclei, and take up one of their stationary states. This is equivalent to writing the wavefunctions as products of electron and nuclear parts, i.e.

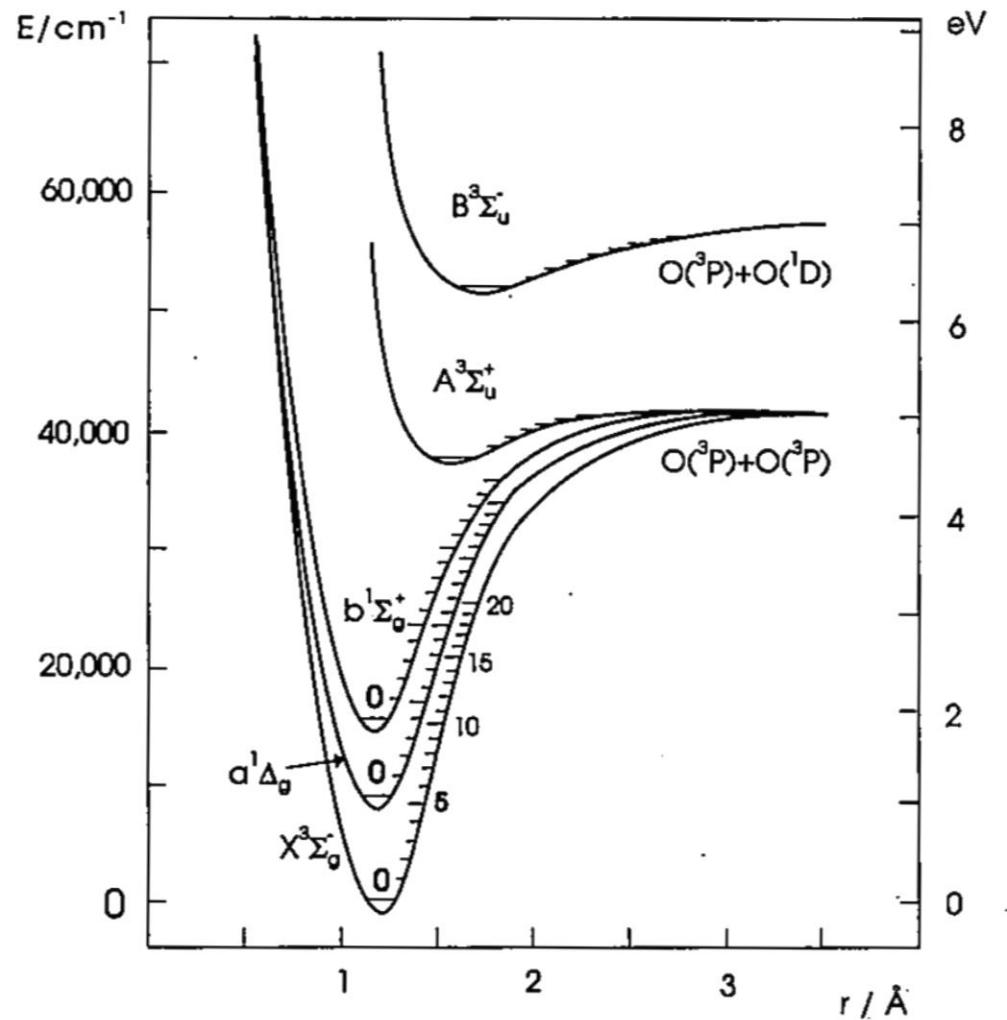
$$\Psi_{evr}(q_e, Q_n) = \sum_i \psi_{ei}(q_e, Q_n) \cdot \psi_{vr}^{(i)}(Q_n)$$

$\psi_e(q_e, Q_n)$ is a solution of the electron Schrödinger equation, for which the Hamiltonian is

$$\hat{H}_e = \frac{1}{2m_e} \sum_e p_e^2 + \hat{V}(q_e, Q_n).$$

Nuclear motion follows the following equation

$$\left(\frac{1}{2} \sum_n \frac{P_n^2}{M_n} + I \right)$$



Degrees of freedom of a molecule

Translation: 3, Rotation: 3 (non-linear molecule), Vibration: $3N-6$
2 (linear molecule) , Vibration: $3N-5$

Normal Modes of water

$\nu_1(a_1)$ 3651 cm^{-1}
symmetric stretch



$\nu_2(a_1)$ 1595 cm^{-1} bend



$\nu_3(b_2)$ 3756 cm^{-1}
antisymmetric stretch

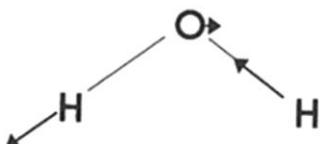


Figure 7.58. The normal vibrational modes of H_2O .

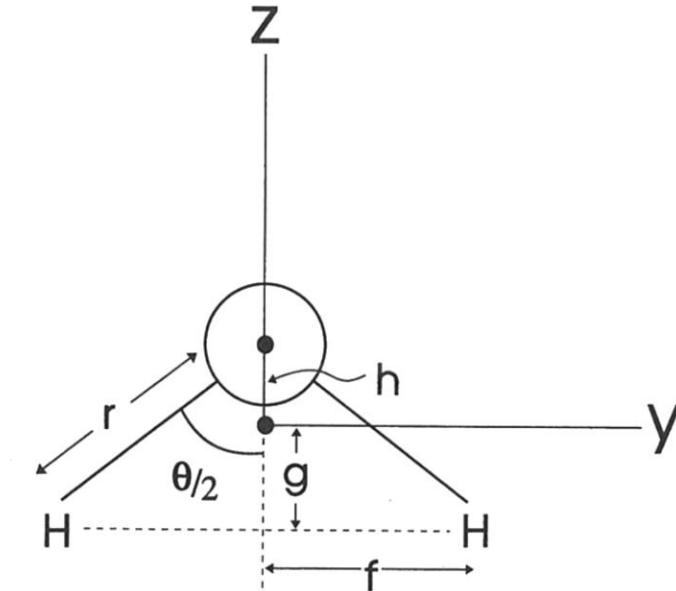


Figure 6.5. The H_2O molecule.

Rotational Spectroscopy

Energy level:

A, B, C :moments of inertia

Diatom and linear molecule

$$G(J) = B\vec{R} \bullet \vec{R} = B(\vec{J} - \vec{L} - \vec{S})^2 = B[J(J+1) - \Omega^2] + DJ^2(J+1)^2$$

Nonlinear molecule

$$A \geq B \geq C$$

B=C or A=B Prolate/Oblate symmetric top

$$G(J, K) = BJ(J+1) - (A-B)K^2$$

A ≠ B ≠ C asymmetric top

Selection Rules

$$\mu \neq 0, \Delta J = 0, \pm 1, \Delta K = 0, \pm 1$$

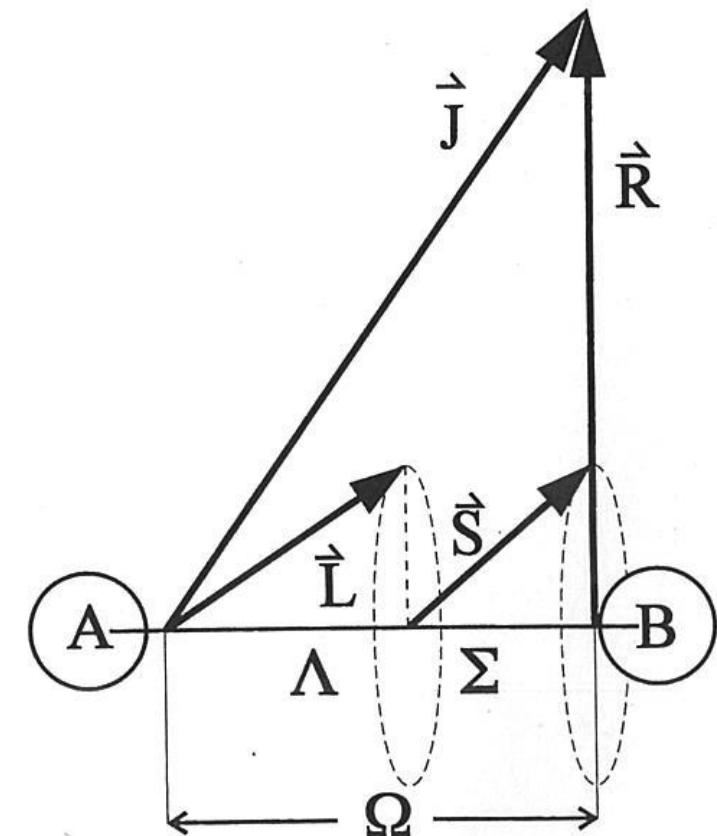


Figure 9.3. Angular momenta in a diatomic molecule.

Vibrational Spectroscopy

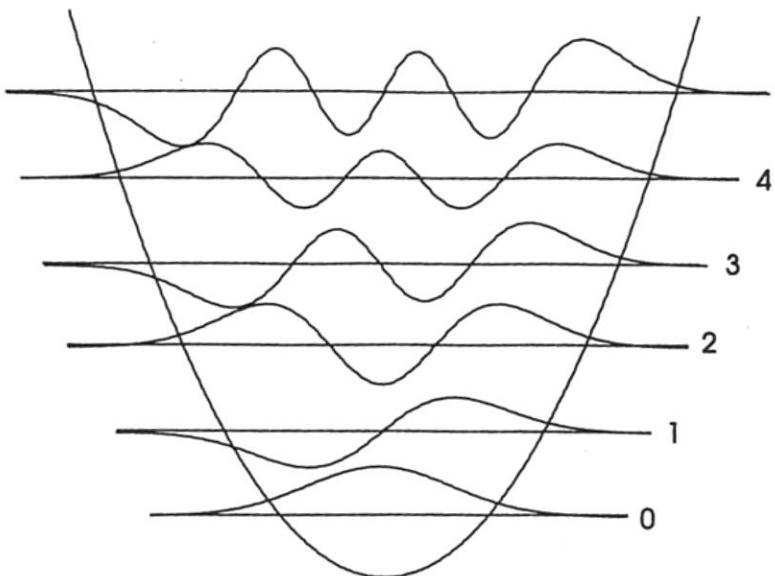


Figure 7.3. The harmonic oscillator wavefunctions.

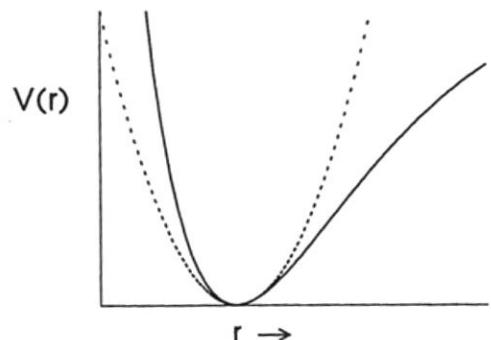
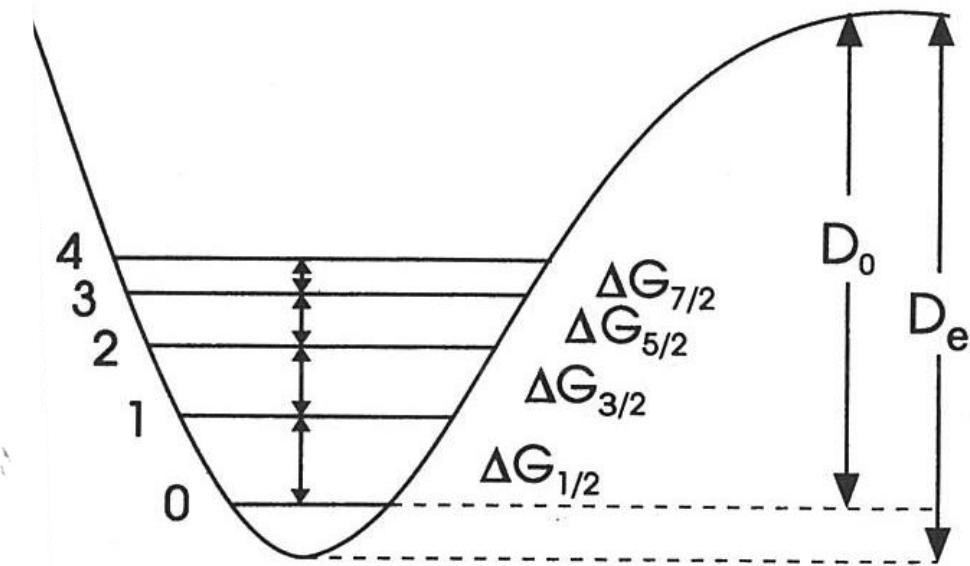


Figure 7.4. A harmonic oscillator potential (dots) as compared to a realistic diatomic potential (solid).



7.8. The vibrational intervals of a diatomic molecule.

Vibrational Energy And Selection Rules

$$\begin{aligned} G(v_1, v_2, \dots, v_i \dots) = & \sum_i^{3N-5(6)} \omega_{ei} \left(v_i + \frac{1}{2} \right) \\ & + \sum_{i \leq j}^{3N-5(6)} \chi_{ij} \left(v_i + \frac{1}{2} \right) \left(v_j + \frac{1}{2} \right) \\ & + \sum_{i \leq j}^{3N-5(6)} g_{ij} I_i I_j \end{aligned}$$

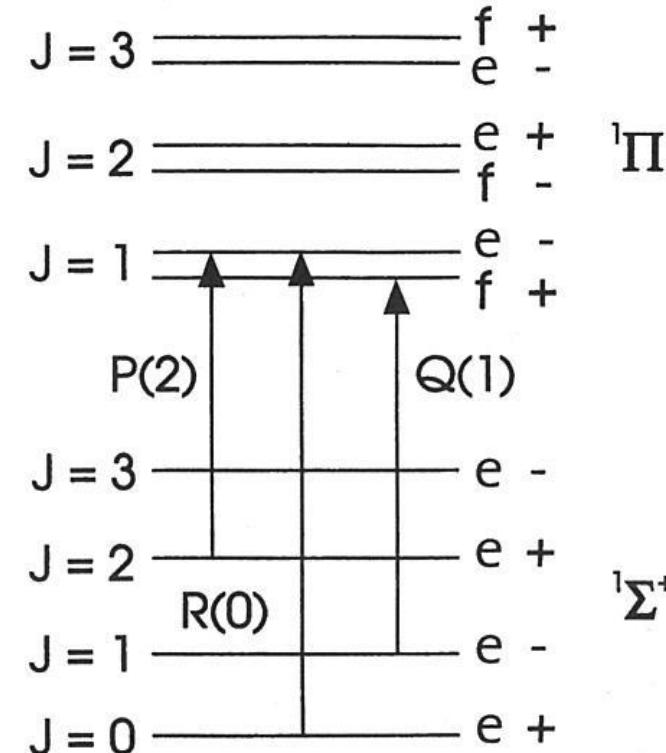
Pure vibration-rotational transition

$\partial\mu/\partial x \neq 0$, μ : dipole moment, x : vibrational coordinate

$$\Delta v_i = \pm 1$$

Selection Rules of Electronic band

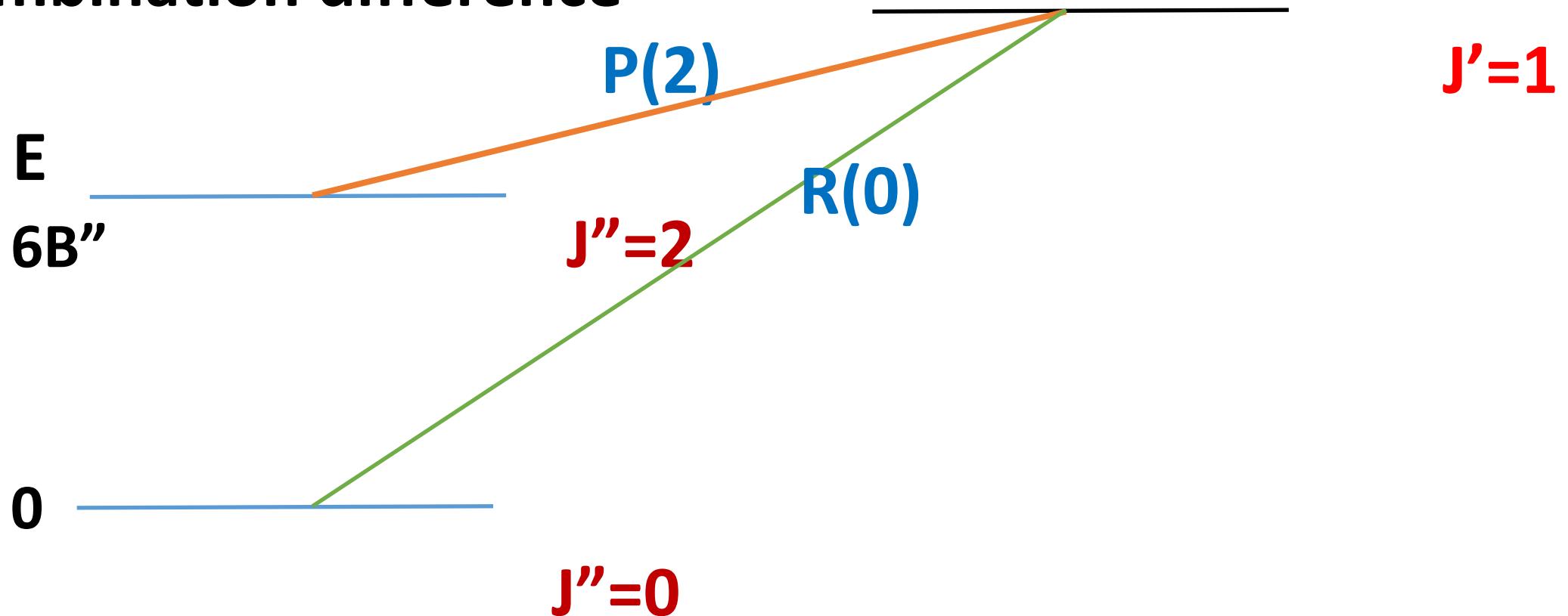
- $\langle f | \hat{\mu} | i \rangle \neq 0$
- $\langle v_f | v_i \rangle \neq 0$
 - Franck-Condon Factor:
 $\langle v_f | v_i \rangle^2$
- **Total Parity**
 $+ \leftrightarrow -$
- $\Delta S = 0$
- $\Sigma^+ \leftrightarrow \Sigma^+, \Sigma^- \leftrightarrow \Sigma^-$
- **Selection rules of angular momentum**



7.33. Energy-level diagram for a $\Pi - \Sigma$ transition.

How to analyze a spectrum?

Combination difference



Basic concepts

Harmonic oscillators and Rigid Rotors

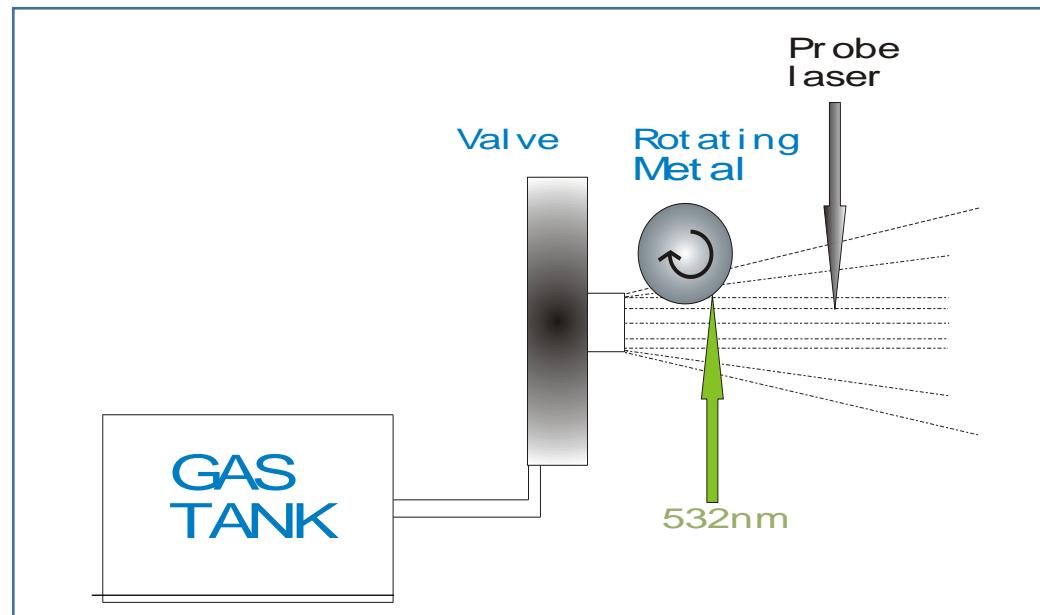
Perturbation theory and effective Hamiltonian

Laser spectroscopic studies of ScC radical in our laboratory

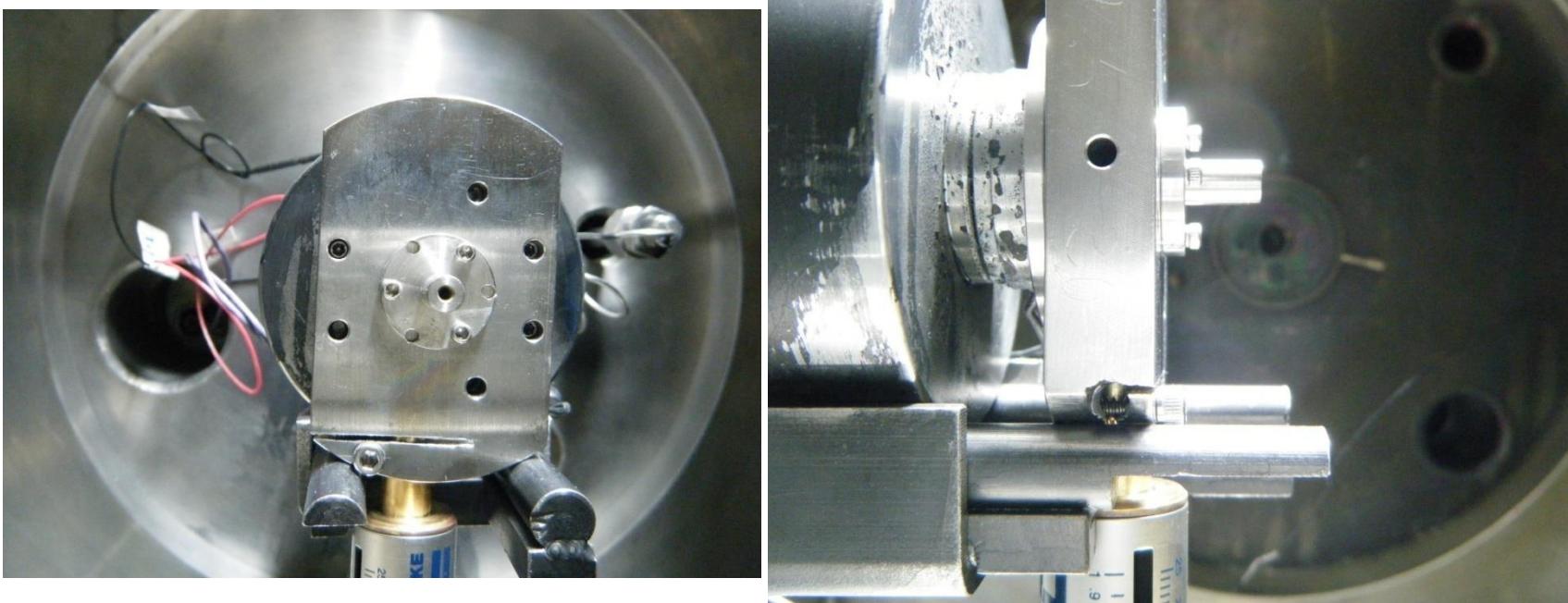
1. C.W. Chen, A. J. Merer, and Yen-Chu Hsu, *J. Chem. Phys.* **149**, 074302-1-074302-15 (2018).
2. C.W. Chen, A. J. Merer, and Yen-Chu Hsu, *J. Mol. Spectrosc.* **361**, 40-46(2019).

Laser Ablation of Metal Rod

Metastable Sc atoms generated by 532nm-laser ablation of a rotating Sc rod were reacted with CH₄ or C₂H₂ seeded in Ar or He under supersonic beam conditions. A Nd:Yag pumped dye laser was used to probe the products by laser-induced.



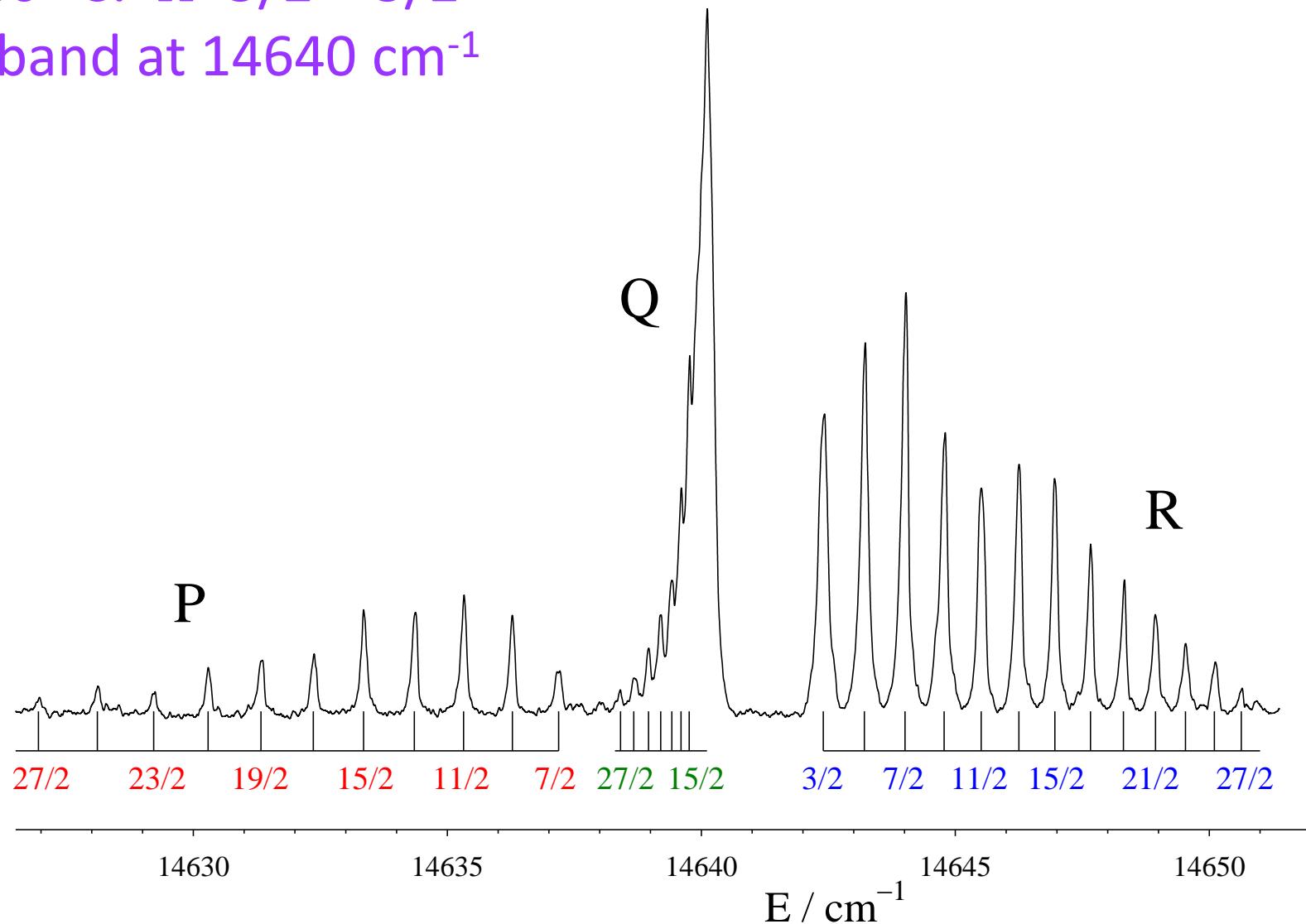
Laser Ablation setup



Introduction of Metal Carbides

- Sc atom is the least abundant of the *3d transition metals* in the Earth's crust.
- *Scandium oxide, ScO*, gives comparatively intense band systems in the spectra of cool M-type stars.
With carbon and oxygen having comparable cosmic abundances, *ScC* is a molecule of possible astrophysical significance.
- *ScC and MnC* are the only remaining *3d transition metals monocarbides that have not been characterized*.

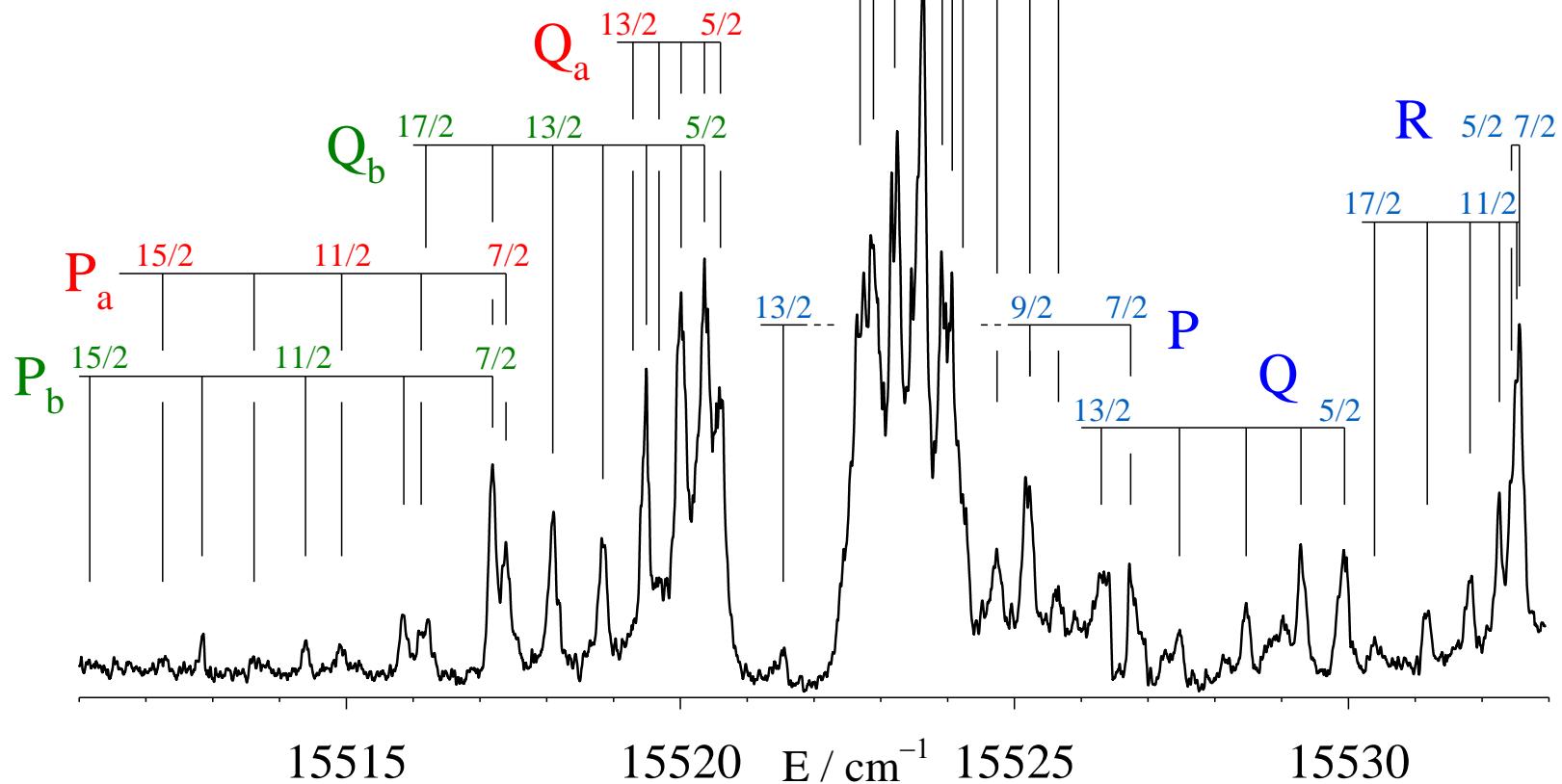
Sc^{13}C : $\Omega=5/2 - 3/2$
band at 14640 cm^{-1}



ScC; the 15521 cm^{-1} bands

$$\Omega = 5/2 - 3/2$$

Upper state Ω -doubling
proportional to $J(J+1)$



Absorption and Emission

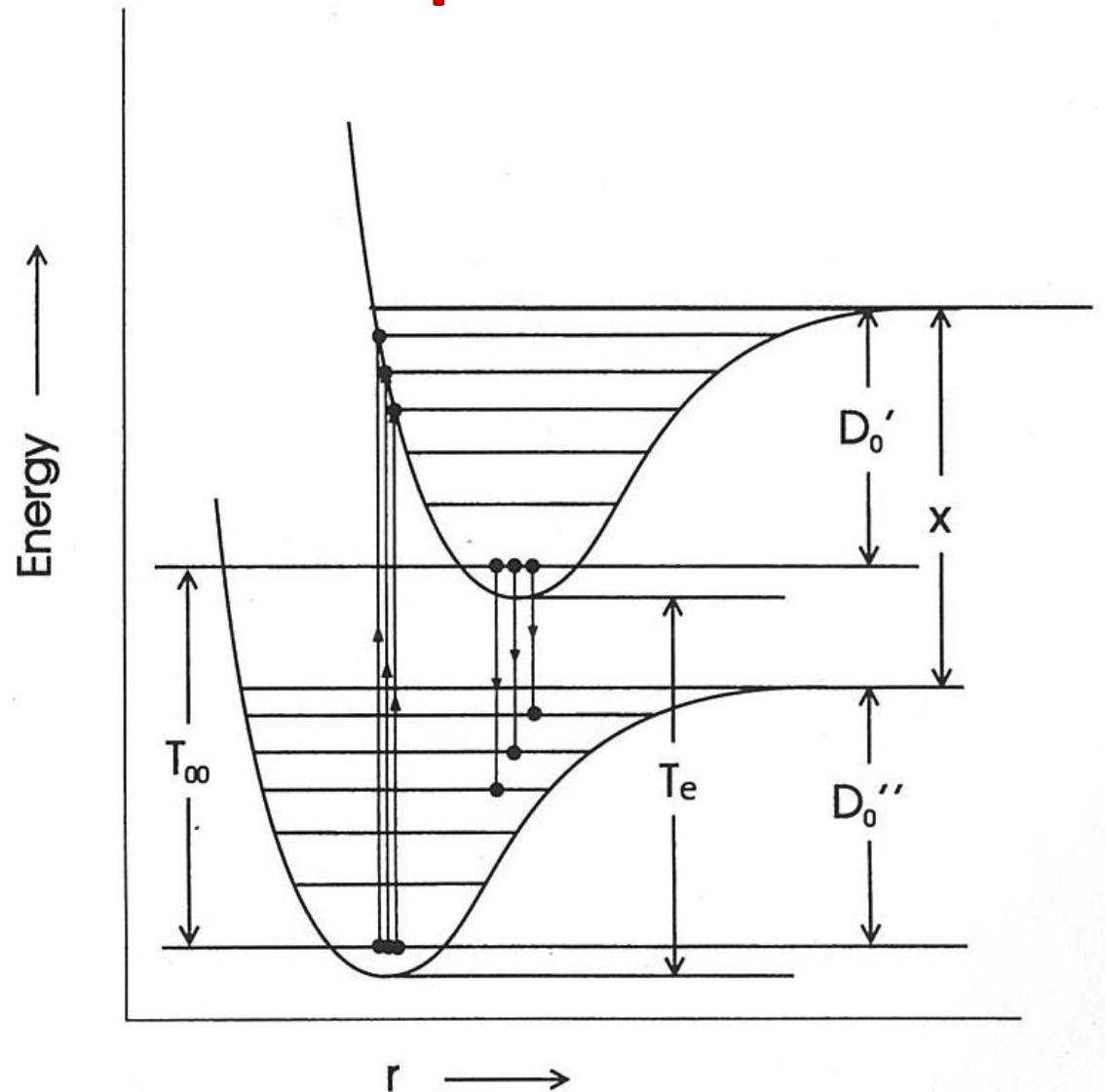
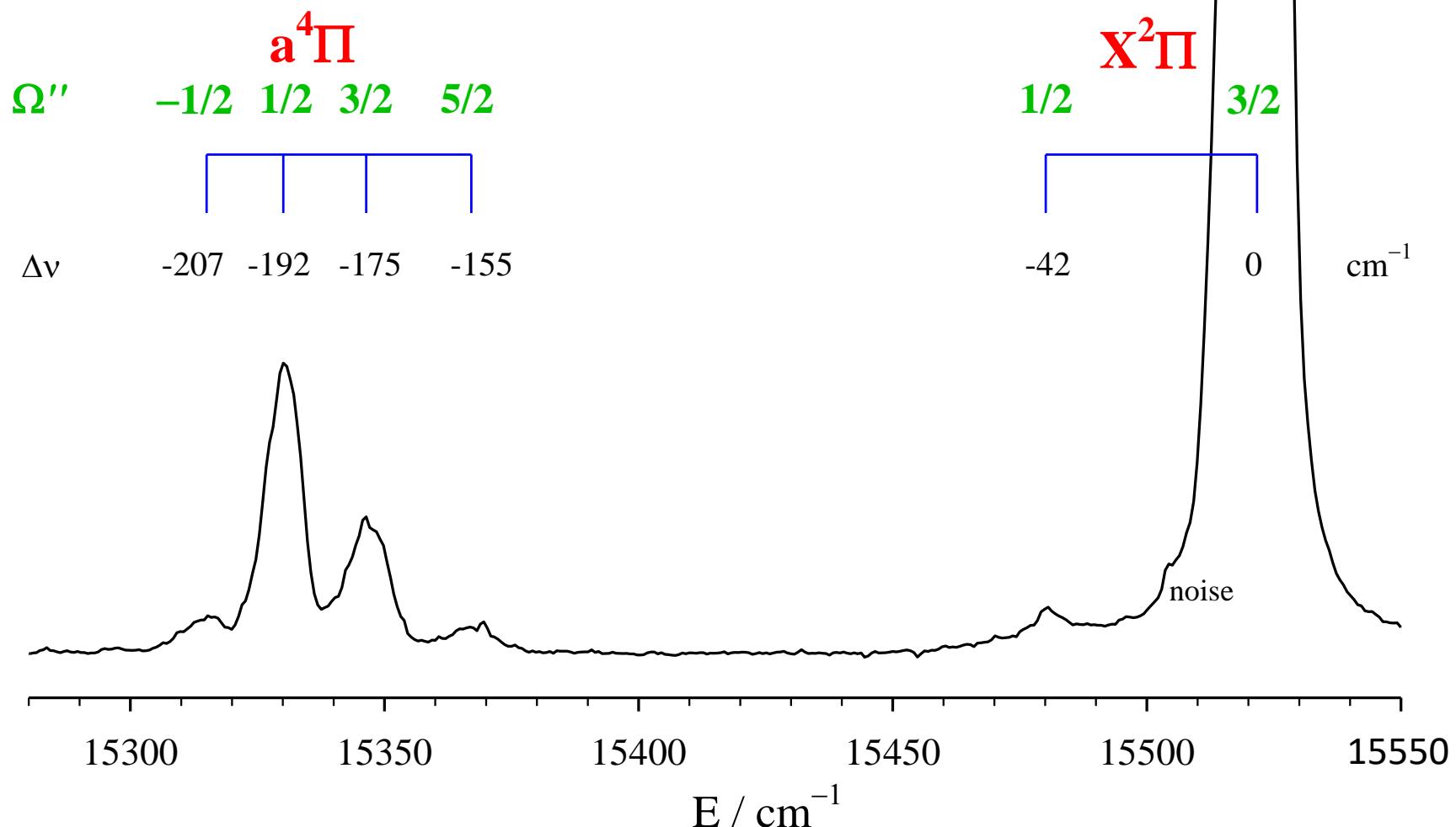


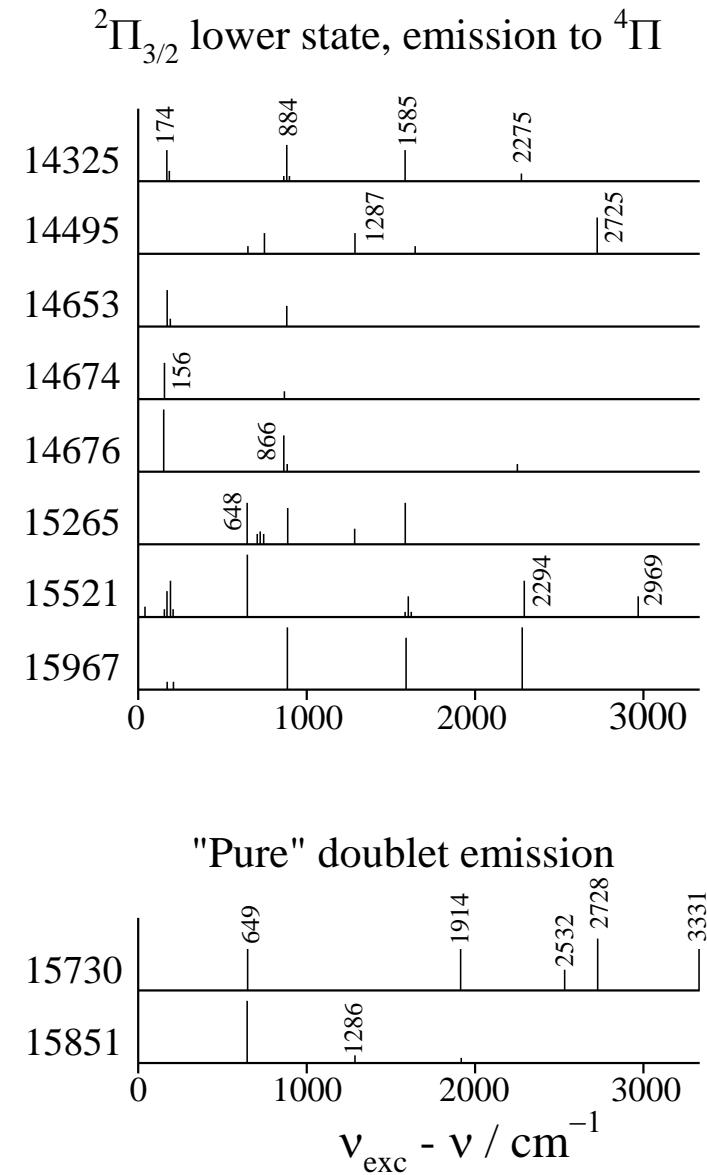
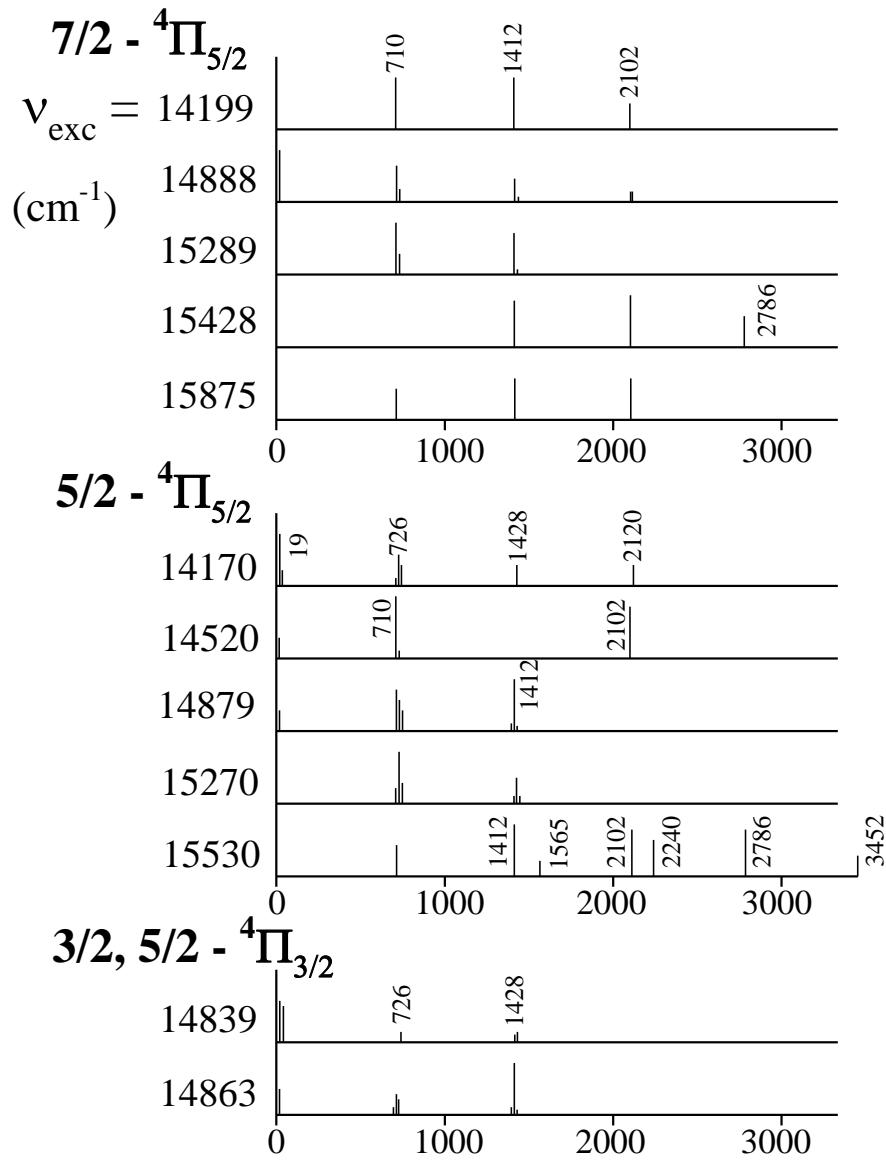
Figure 9.27. Absorption and emission progressions.

Emission from the $\Omega=5/2 - X^2\Pi_{3/2}$ band at 15521 cm^{-1}

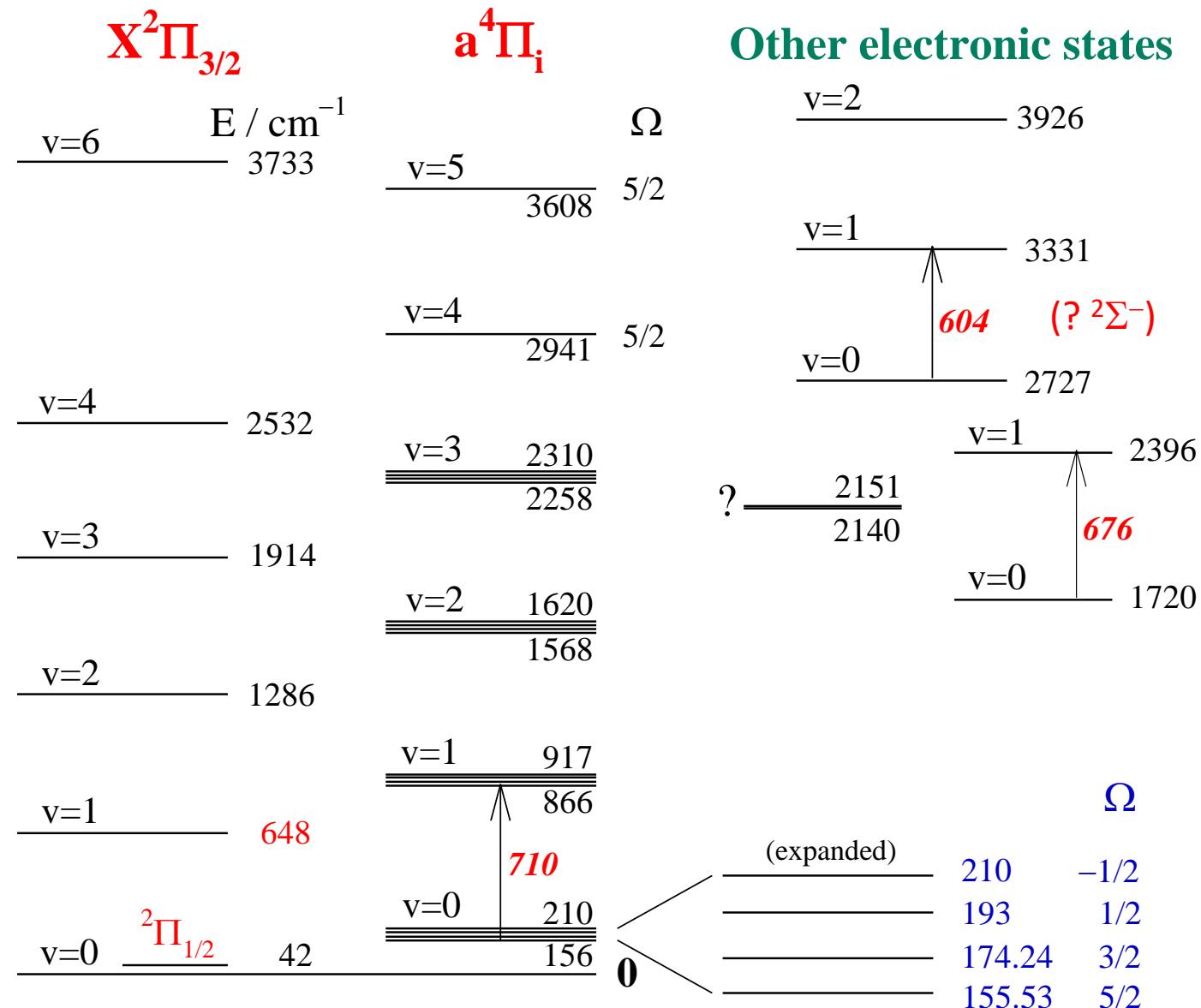
Sc¹²C



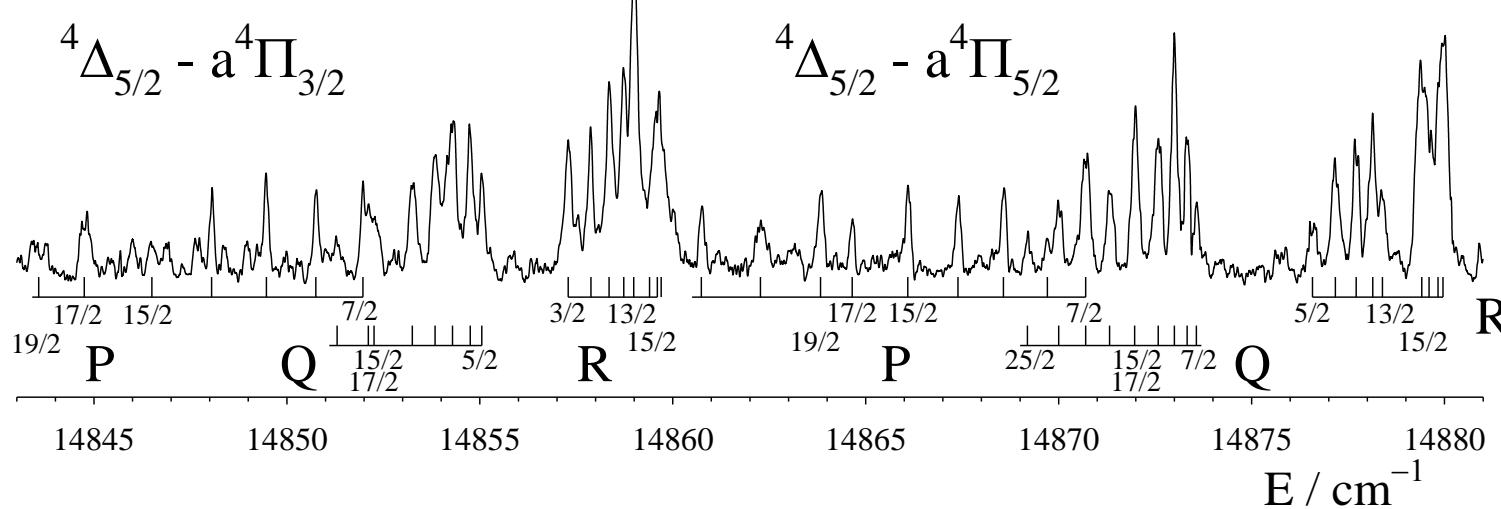
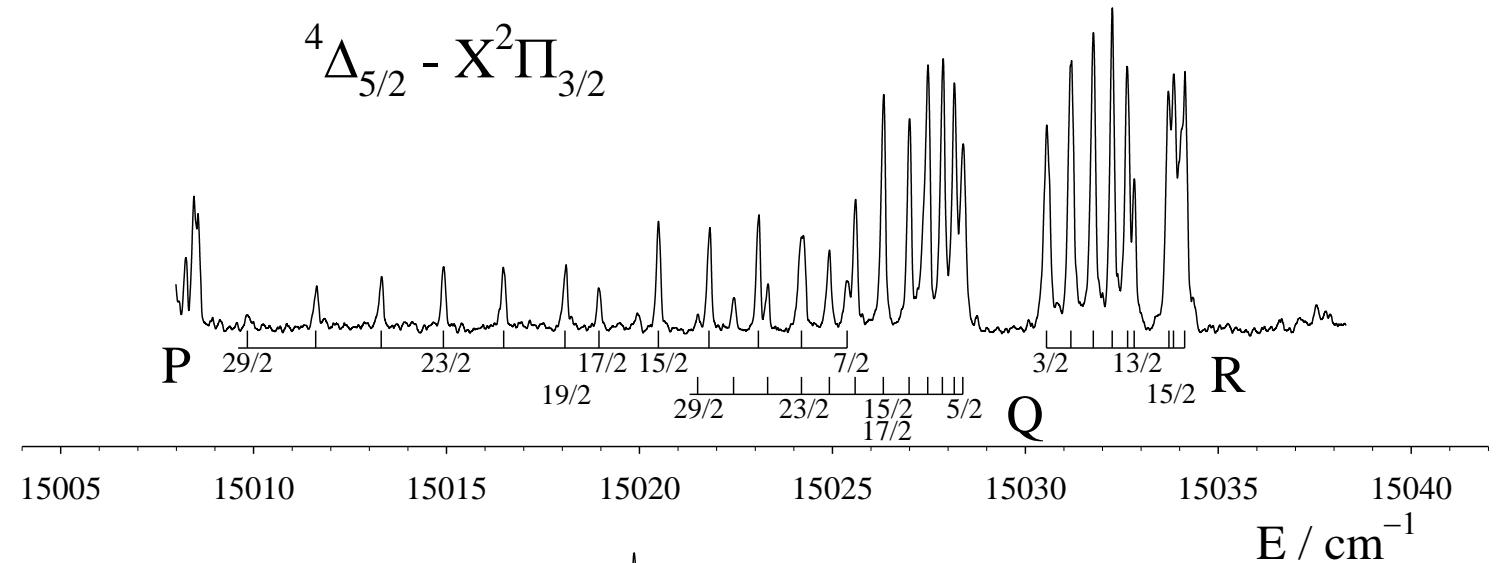
Emission spectra from selected bands of Sc^{12}C



Sc¹²C: low-lying spin and vibrational levels



Sc^{13}C : perturbation in the 15028 cm^{-1} level



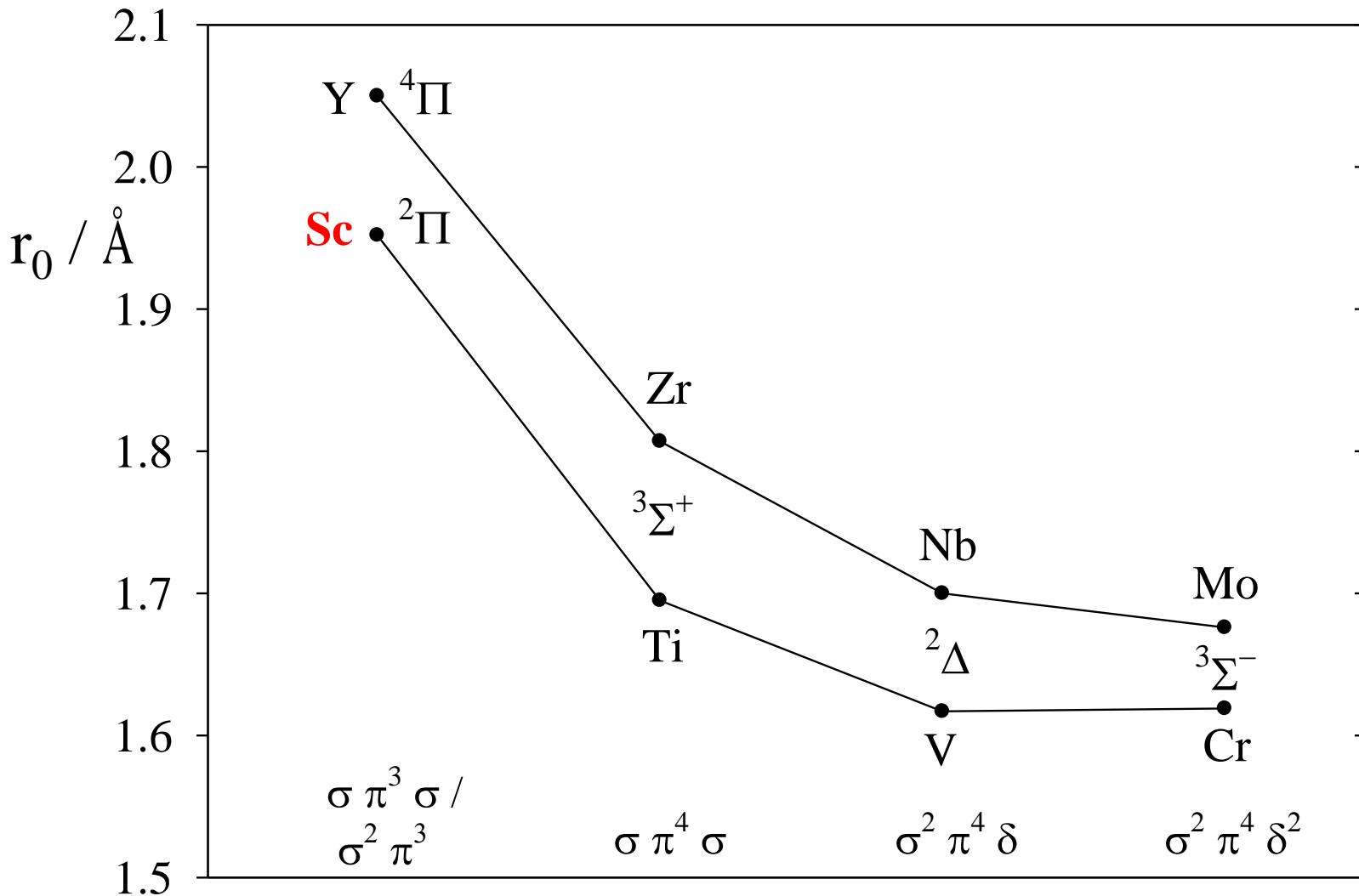
**Rotational constants for the $X^2\Pi$, $a^4\Pi_{5/2}$ and $a^4\Pi_{3/2}$ states of
Sc¹²C and Sc¹³C. Values in cm⁻¹. Quoted error limits are 3σ .**

		Sc ¹² C	Sc ¹³ C
$X^2\Pi_{3/2}$	T ₀	0	0
	B	$0.4616_7 \pm 0.0003_9$	$0.4341_7 \pm 0.0004_0$
	10^7D	9.4 fixed	8.2 fixed
$X^2\Pi_{1/2}$	T ₀	$39.46_9 \pm 0.01_8$	$39.31_8 \pm 0.02_3$
	B	$0.4724_5 \pm 0.0012_3$	$0.4418_4 \pm 0.0011_5$
	10^7D	0 fixed	0 fixed
$a^4\Pi_{5/2}$	T ₀	$155.58_4 \pm 0.02_6$	$154.71_7 \pm 0.02_8$
	B	$0.4534_7 \pm 0.0009_5$	$0.4278_9 \pm 0.0009_5$
	10^4D	$-0.19_8 \pm 0.07_0$	$-0.15_0 \pm 0.08_0$
$a^4\Pi_{3/2}$	T ₀	$174.31_1 \pm 0.03_3$	$173.24_2 \pm 0.03_6$
	B	$0.4703_9 \pm 0.0018_6$	$0.4447_0 \pm 0.0022_2$
	10^4D	$-0.34_3 \pm 0.21_6$	$-0.20_8 \pm 0.30_4$
	σ	0.017	0.018

$$r_0 = 1.952(1) \text{ \AA}$$

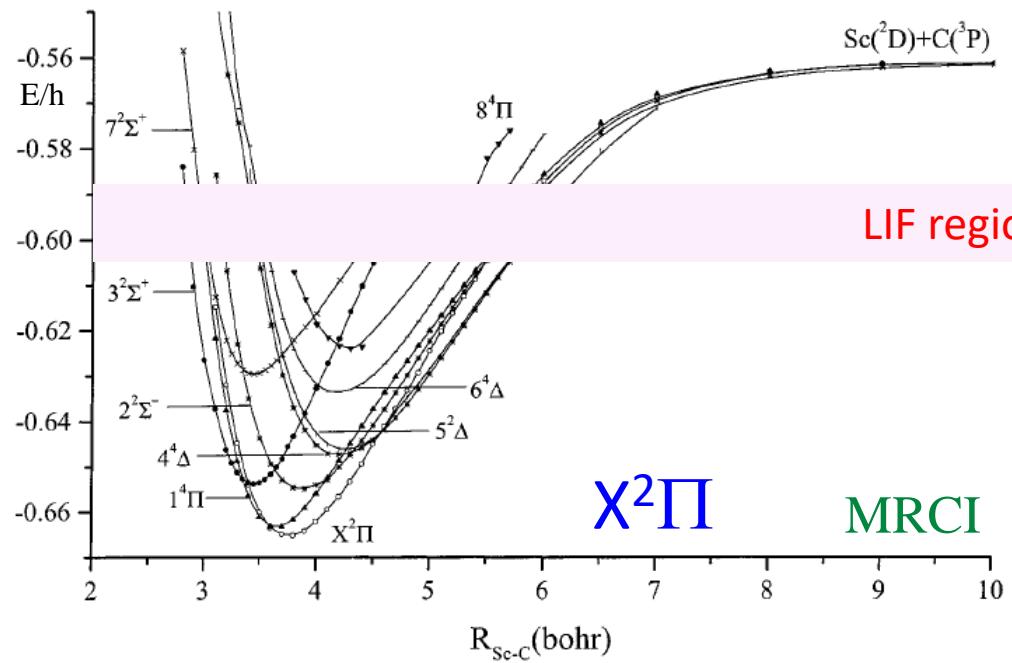
$$r_0 = 1.928(7) \text{ \AA}$$

Bond lengths in the “early” 3d and 4d monocarbides

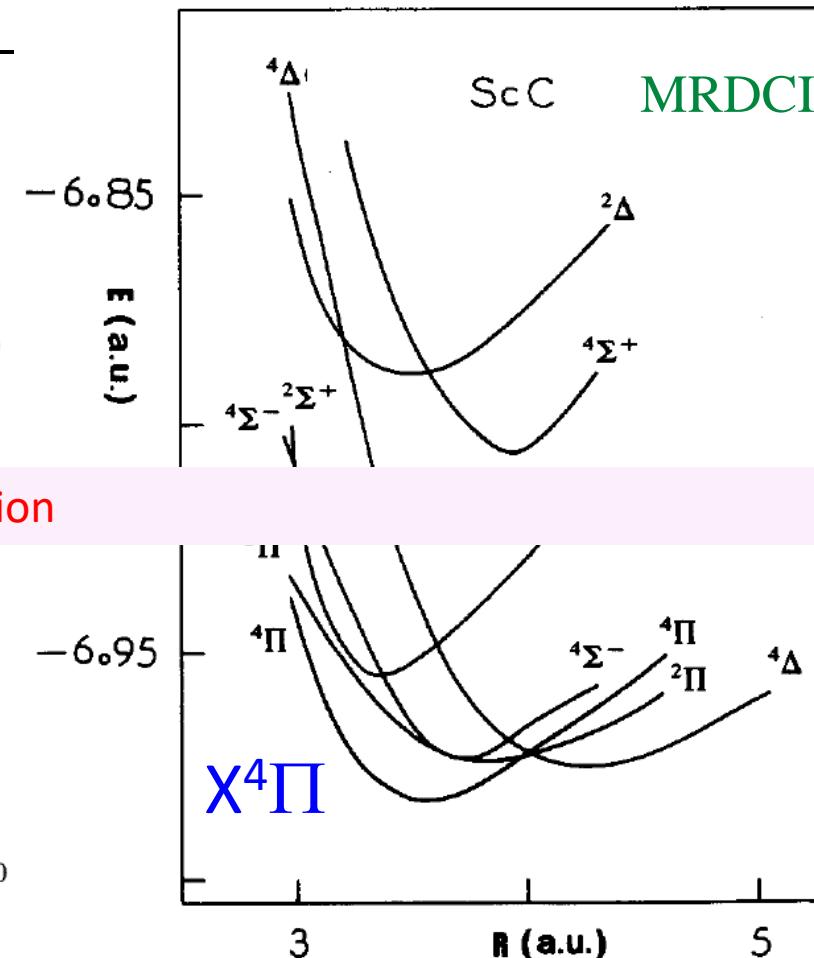


Potential energy curves for ScC

Two *ab initio* calculations have been done:-



A. Kalemos, A. Mavridis and J.F. Harrison,
J. Phys. Chem. A, **105**, 755 (2001)



G.H. Jeung and J. Koutecký,
J. Chem. Phys. **88**, 3747 (1988)

The low-lying electronic states of ScC

The lowest observed state has $\Omega = 3/2$, consistent with $^2\Pi_i$ or $^2\Delta_r$. However no low-lying $^2\Delta_r$ states are expected.

We find it has

$$r_0 = 1.952(1) \text{ \AA} \quad \Delta G_{1/2} = 648 \text{ cm}^{-1}$$

Kalemos *et al* calculate

$$r_e = 1.973 \text{ \AA} \quad \omega_e = 690 \text{ cm}^{-1}$$

A very low-lying $^4\Pi_i$ state has $A = 18.7 \text{ cm}^{-1}$ (spin-orbit coupling).

We find

$$r_0 = 1.928(7) \text{ \AA} ; \Delta G_{1/2} = 710 \text{ cm}^{-1}; T_0 = 155 \text{ cm}^{-1}$$

Kalemos *et al*:

$$r_e = 1.937 \text{ \AA} ; \quad \omega_e = 694 \text{ cm}^{-1}; \quad T_e = 432 \text{ cm}^{-1}$$

The

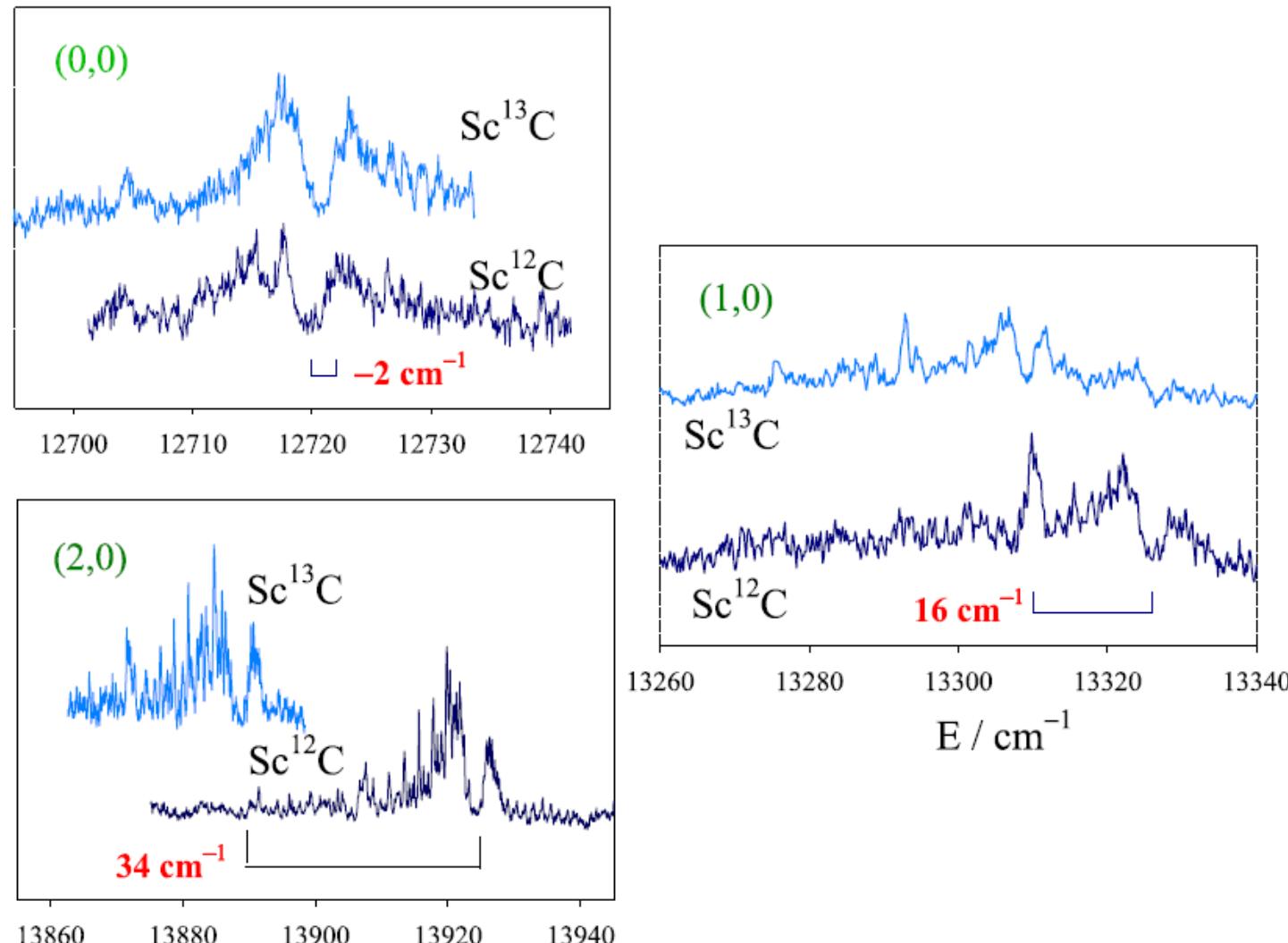


Fig. 1. The (0,0), (1,0) and (2,0) bands of the ${}^4\Sigma - {}^4\Pi$ transitions of Sc¹²C and Sc¹³C, illustrating their isotope shifts. The brackets point to the ${}^4\Sigma - {}^4\Pi_{5/2}$ sub-band origins, which are prominent dips in the rotational structure.

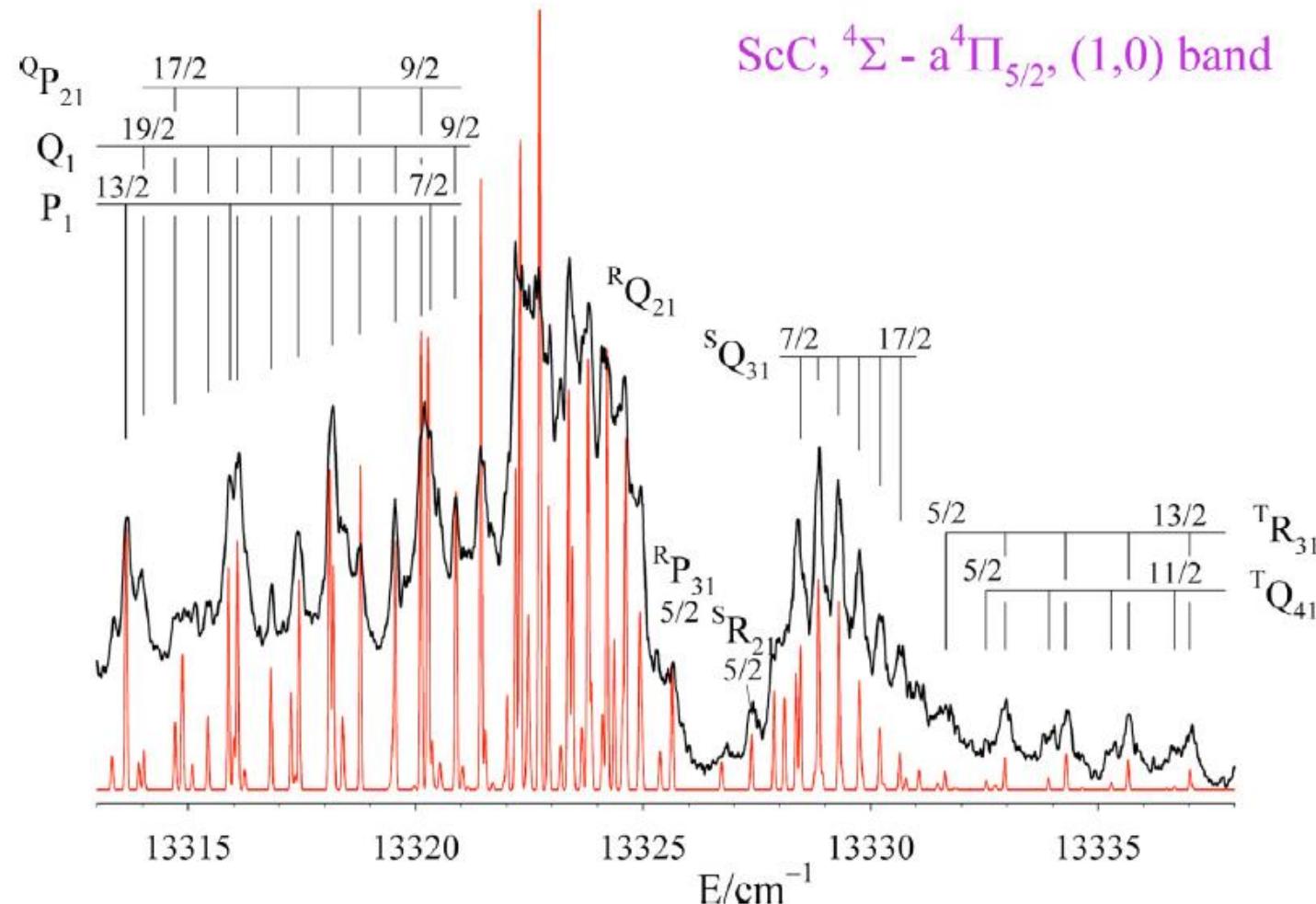


Fig. 2. The Sc^{12}C , $^4\Sigma - \text{a}^4\Pi_{5/2}, (1,0)$ sub-band, with the assignments of six of the 12 branches marked. The band contour calculated from the final least squares results is also shown (colored red in the on-line version). The line width for the calculated contour is 0.1 cm^{-1} , which is narrower than the experimental line width in order to show the details of the rotational structure. A temperature of 16 K was assumed. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

Table 1Matrix elements for the spin and rotation of a ${}^4\Sigma^-$ electronic state.

	$ J, 1/2 e/f\rangle$	$ J, 3/2 e/f\rangle$
$\langle J, 1/2 e/f $	$T_0 - 2\lambda - 7\gamma/2 + B(x+4)$ $- D [(x+4)^2 + 7x + 4]$ $\mp (2J+1)[B - \gamma/2 - 2D(x+4)]$	symmetrical
$ J, 3/2 e/f $	$-\sqrt{3x}[B - \gamma/2 - 2D(x+2)$ $\pm 2D(J+1/2)]$	$T_0 + 2\lambda - 3\gamma/2 + Bx$ $- D(x^2 + 3x)$

$$x = J(J+1) - 3/4.$$

For a ${}^4\Sigma^-$ state the F_1 and F_3 spin component levels correspond to the sum basis $2^{-1/2} [|J, \Omega\rangle + |J, -\Omega\rangle]$, and have e symmetry; the F_2 and F_4 levels (difference basis, $2^{-1/2} [|J, \Omega\rangle - |J, -\Omega\rangle]$) have f symmetry. The parameters λ and γ are the coefficients of the spin-spin and spin-rotation interactions, respectively.

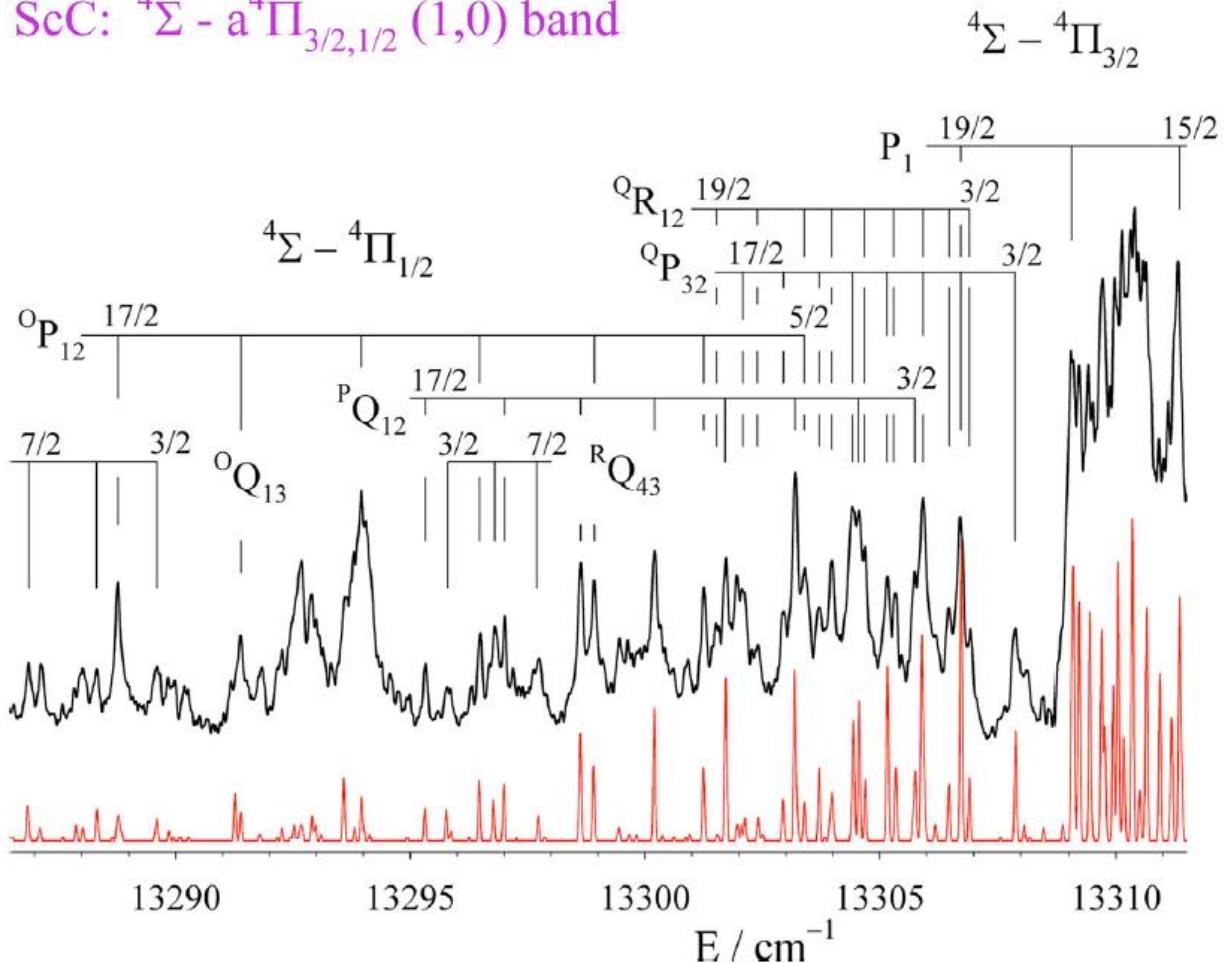
Table 2Matrix elements for the spin and rotation of a $^4\Pi$ electronic state.

	$ J, -1/2 e/f\rangle$	$ J, 1/2 e/f\rangle$	$ J, 3/2 e/f\rangle$	$ J, 5/2 e/f\rangle$
$\langle J, -1/2 e/f $	$T_0 - 3A/2 + 2\lambda + (z + 1)$ $(B - 3A_D/2) - D(z^2 + 5z + 1)$			
$\langle J, 1/2 e/f $	$-\sqrt{3}(J + 1/2)[B - A_D - 2D(z + 2)] \mp \sqrt{3}(o + p + q)$	$T_0 - A/2 - 2\lambda + (z + 3)(B - A_D/2)$ $- D(z^2 + 13z + 5) \pm (J + 1/2)(p + 2q)$	symmetrical	
$\langle J, 3/2 e/f $	$-\sqrt{3}(z - 1)[2D(J + 1/2) \mp (1/2)(p + 2q)]$	$-2\sqrt{z - 1}[B - 2D(z + 2) \pm (1/4)q(J + 1/2)]$	$T_0 + A/2 - 2\lambda + (z + 1)(B + A_D/2)$ $- D(z^2 + 9z - 15)$	
$\langle J, 5/2 e/f $	$\mp (q/2)\sqrt{(z - 1)(z - 4)}$	$-2D\sqrt{3(z - 1)}\sqrt{(z - 4)}$	$-\sqrt{3(z - 4)}[B + A_D - 2D(z - 2)]$	$T_0 + 3A/2 + 2\lambda + (z - 5)(B + 3A_D/2)$ $- D(z^2 - 7z + 13)$

The centrifugal distortions of the rotation (parameter D) and the spin-orbit coupling (parameter A_D) were floated in the first fits, but were not determinable; they were omitted in later fits. The parameters q, p + 2q and o + p + q model the Λ -doubling.

$$z = (J + 1/2)^2.$$

ScC: $^4\Sigma$ - $a^4\Pi_{3/2,1/2}$ (1,0) band



Prospective

1. CRD spectrum of ScC in this region to determine its oscillator strength.
2. measure the hyperfine constants and dipole moments of this band

Fig. 3. The Sc^{12}C , $^4\Sigma - a^4\Pi_{3/2,1/2}$, (1,0) sub-bands, with the assignments of some of the strongest branches marked. The two blue-degraded branches (assignments not marked) forming heads near $13,310 \text{ cm}^{-1}$ are $^R\text{Q}_{32}$ and $^R\text{P}_{42}$. The final band contour is also shown, with conditions as in Fig. 2.

Laser Spectroscopy of C₃ Radical

1. G. Zhang, K.-S. Chen, A. J. Merer, Y.-C. Hsu, W.-J. Chen, S. Sadisavan, and Y.-A. Liao, *J. Chem. Phys.* **122**, 244308(2005).
2. C.-W. Chen, A. J. Merer, J.-M. Chao and Y.-C. Hsu, *J. Mol. Spectrosc.* **263**, 56-70(2010).
3. K.-S. Chen, G. Zhang , A. J. Merer, Y.-C. Hsu, and W.-J. Chen, *J. Mol. Spectrosc.* **267**, 169-171 (2011).
4. Y.-J. Wang, C.-W. Chen, L.-Z. Zhou, A. J. Merer, and Y.-C. Hsu, *J. Phys. Chem. A* **117**, 13878(2013).

Introduction

- The spectrum of the C_3 , $\tilde{A}^1\Pi_u - \tilde{X}^1\Sigma_g^+$ system was first observed in comets by Huggins in 1882.
- Its first laboratory study was reported in 1942 by Herzberg and his co-workers. Since then many studies of the comet system of C_3 have been carried out.
 - 1965, Gausset *et al.*, vibrational and rotational analysis of both the \tilde{A} and \tilde{X} states.
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 - 1998, Izuha and Yamanouchi, double minima found in the \tilde{A} state q_3 coordinate.
 - 2003, McCall *et al.*, reassignment of the R(0) line of the 000-000 band.
 - 2005, Zhang *et al.*, perturbations of the \tilde{A} , 000 state have been observed and analyzed.
 - 2010, Chen *et al.*, extended the study of the \tilde{A} state up to 5000 cm^{-1} above its zero point energy and found a strong anharmonic resonance between v_1 and v_3 .

Gerhard Herzberg

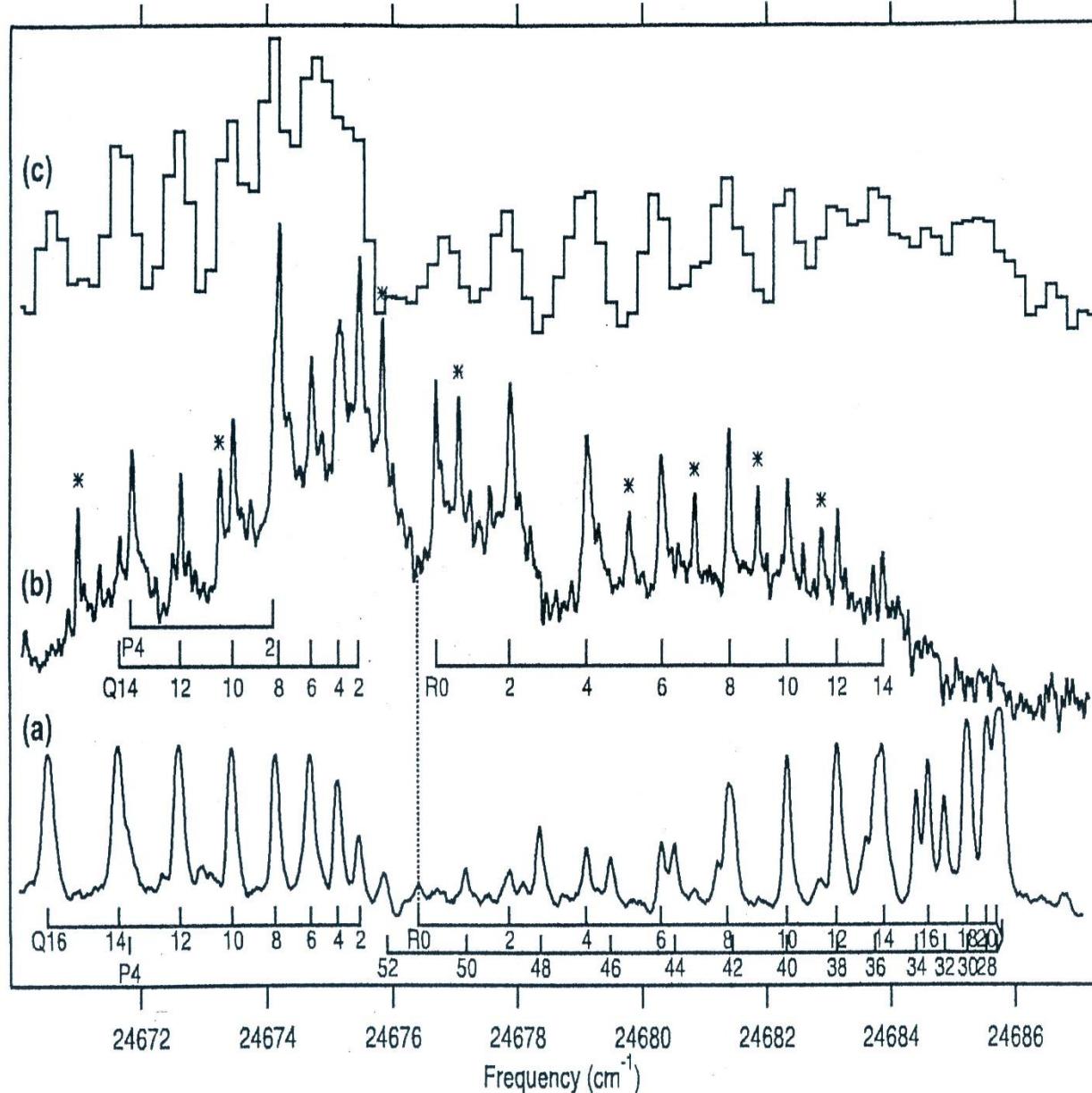
- December 1904- March 1999
- Nobel Prize in Chemistry, 1971
- Contributions to the field of Atomic and Molecular Spectroscopy
- Books:
 1. Atomic Spectra and Atomic Structure, 1937
 2. The spectra and structures of simple free radicals : an introduction to molecular spectroscopy.
 3. Molecular Spectra and Molecular Structure – I. Spectra of Diatomic Molecules
 4. Molecular Spectra and Molecular Structure –II. Infrared and Raman Spectra of Polyatomic Molecules
 5. Molecular Spectra and Molecular Structure: III. Electronic Spectra and Electronic Structure of Polyatomic Molecules
 6. Molecular Spectra and Molecular Structure: IV. Constants of Diatomic Molecules

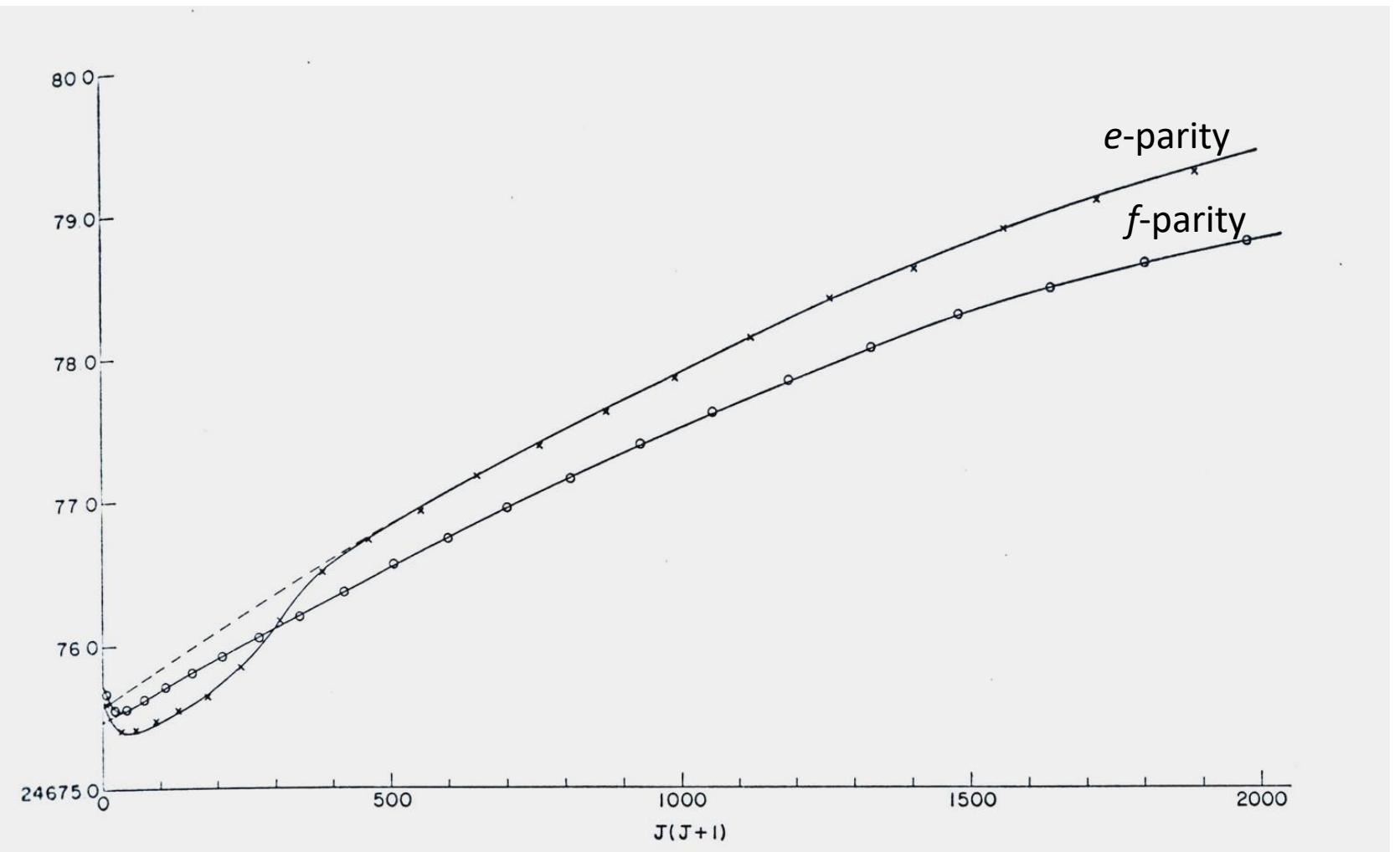


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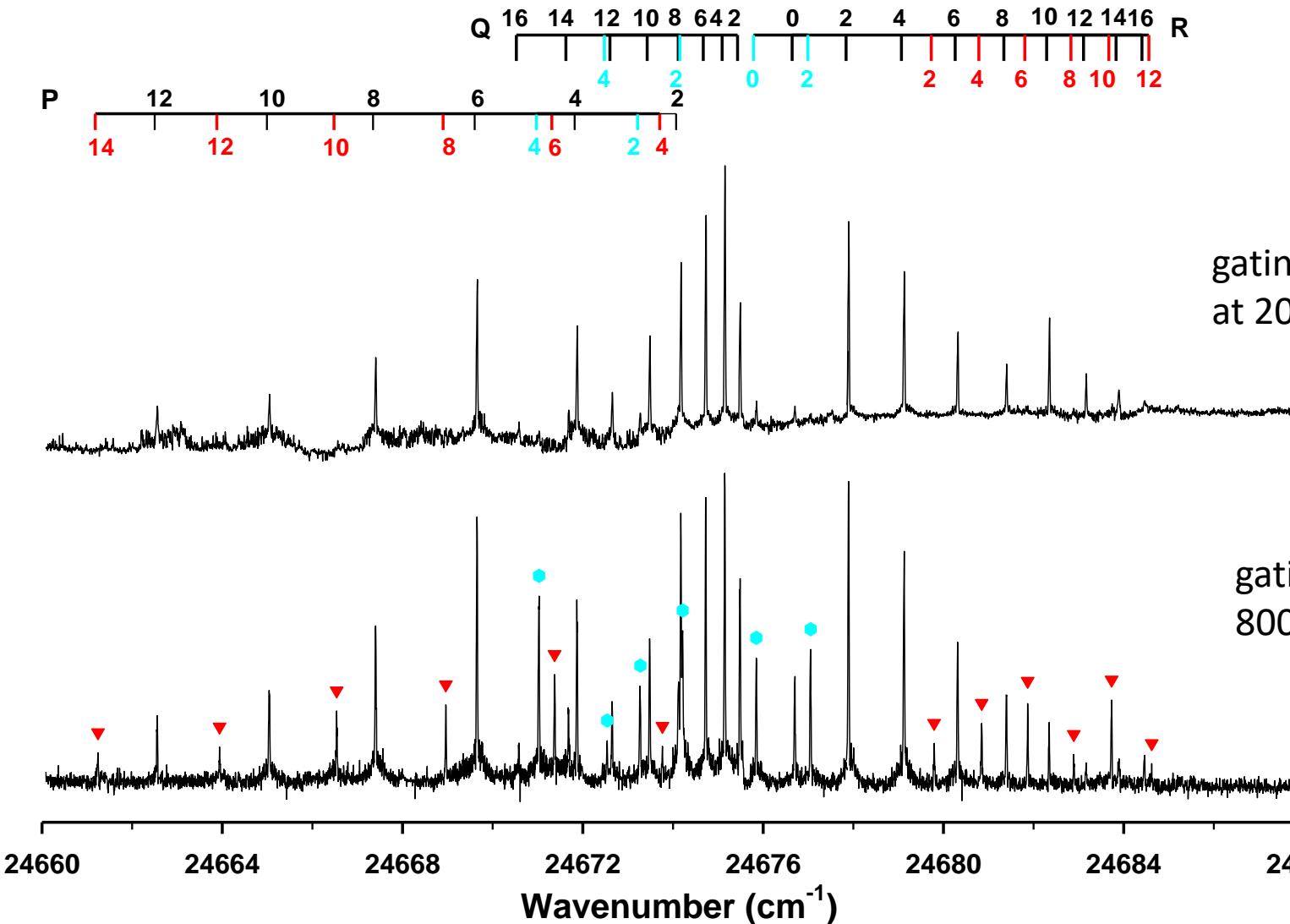
- (a) Gausset et al, *Astrophys. J.* 142, 45(1965)
Flash photolysis of the mixture of diazomethane and inert gas
- (b) Cavity-ring down Spectroscopy of C_3
- (c) HD 62542





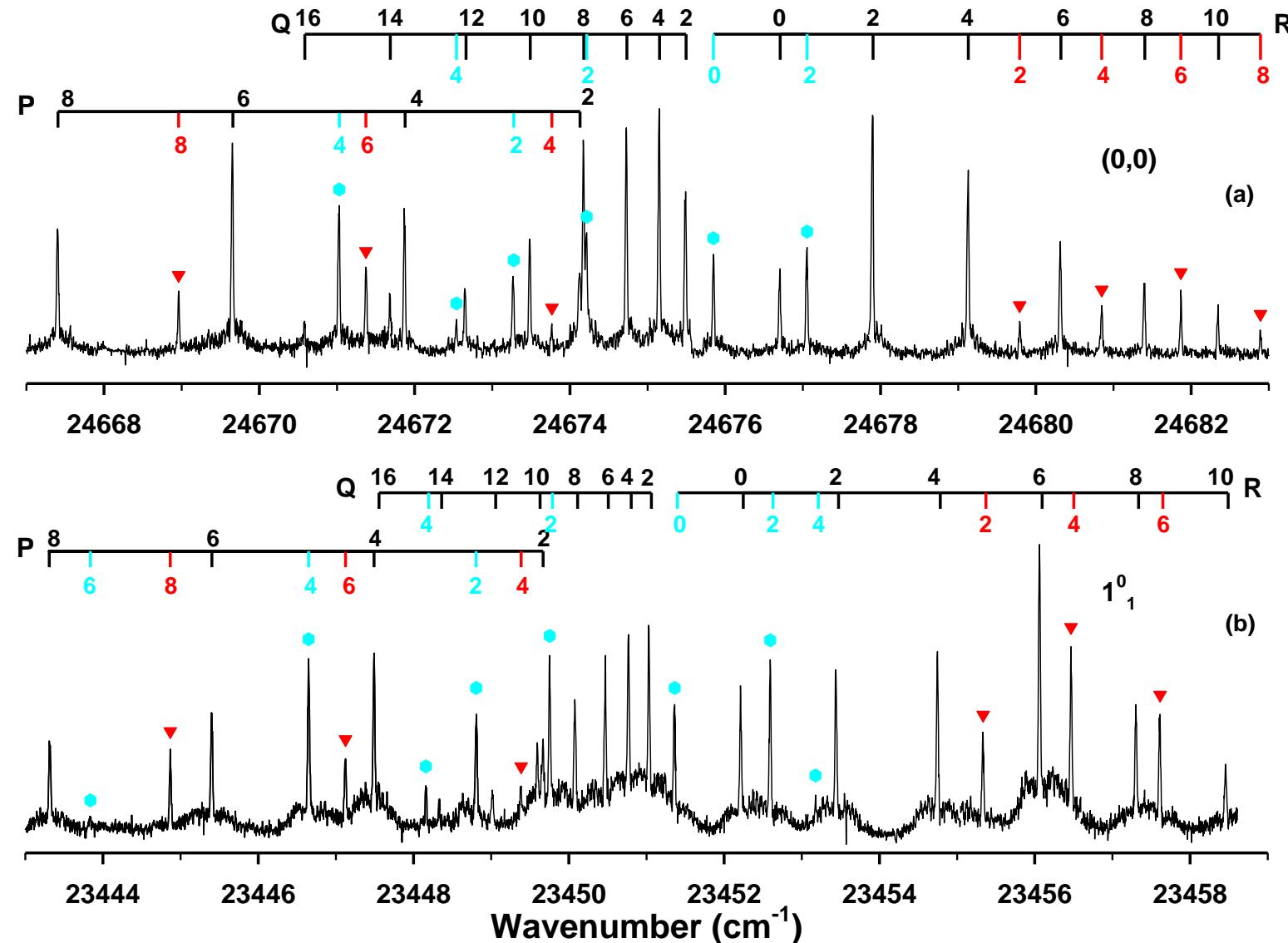
L. Gausset et. al., *Astrophys. J.* **42**, 45(1965)

Fluorescence Excitation Spectra of the 000-000 band of the $\tilde{A} - \tilde{X}$ system of C_3



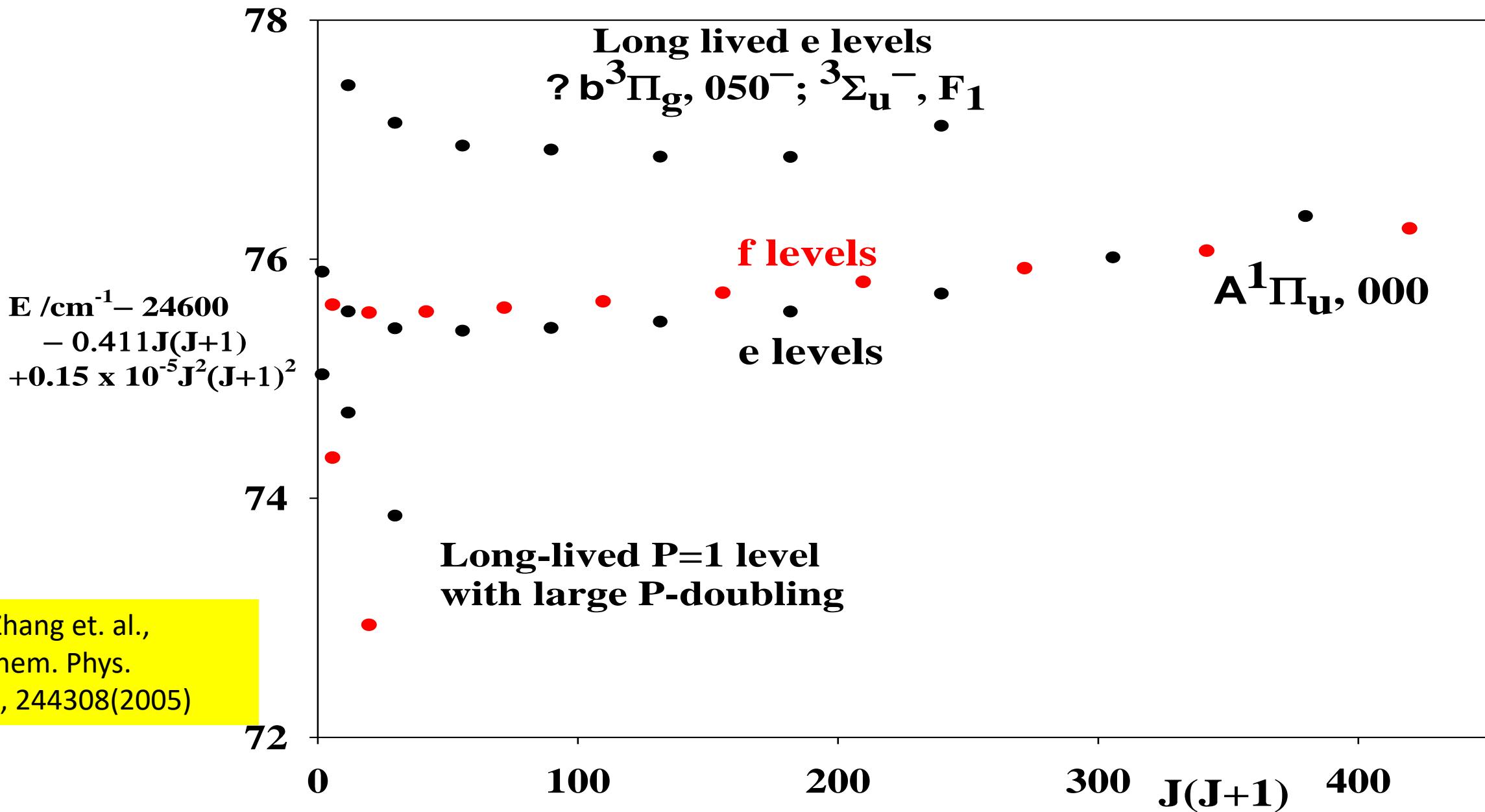
G. Zhang et. al.,
J. Chem. Phys.
122, 244308(2005)

Fluorescence Excitation Spectra of the (0,0) and 1^0_1 bands of C₃



gating at
800-2300 ns

G. Zhang et. al.,
J. Chem. Phys.
122, 244308(2005)



Measured lifetimes for some low-J rotational levels of the $\tilde{\Lambda}^1\Pi_u$, 000 state of C₃ (Values in parentheses represent three standard deviation of the fit.)

Single exponential decays (Main band)

Line	Wavenumber/cm ⁻¹	τ/ns
R(2)	24677.907	227(6)
R(4)	24679.134	214(15)
R(6)	24680.329	212(2)
Q(2)	24675.498	209(6)
Q(4)	24675.158	195(9)
Q(6)	24674.736	192(6)
P(4)	24671.878	227(6)

Bi-exponential decays (Perturbing states)

Line	Wavenumber/cm ⁻¹	τ_1/ns	τ_2/ns
P(2)	24673.270	237(4)	470(15)
P(4)	24671.032	786(5)	202(5)
R(2)	24677.058	752(51)	216(26)
R(2)	24679.798	223(10)	2190(600)

G. Zhang et. al.,
J. Chem. Phys.
122, 244308(2005)

**Rotational constants derived by least squares from our data for the
 $\tilde{\text{A}}^{-1}\Pi_u$, 000 state of C_3 and its perturbing states. Values in cm^{-1} .**

T_0^{Π}	24675.632	\pm	0.080	T_0^{Σ}	24682.12	\pm	0.74
B^{Π}	0.41261	0.00030		B^{Σ}	0.4411	0.0032	
$10^6 D^{\Pi}$	0.298 (fixed)			λ	-7.22	1.32	
q^{Π}	-0.00125	0.00059		T_0^P	24675.68	0.14	
ξ	0.744	0.063		B^P	0.3154	0.0072	
a	0.436	0.060		q^P	-0.076	0.007	

r.m.s. error **0.0366**

Schmidt et. al,
MNRAS 441,
1131(2014)

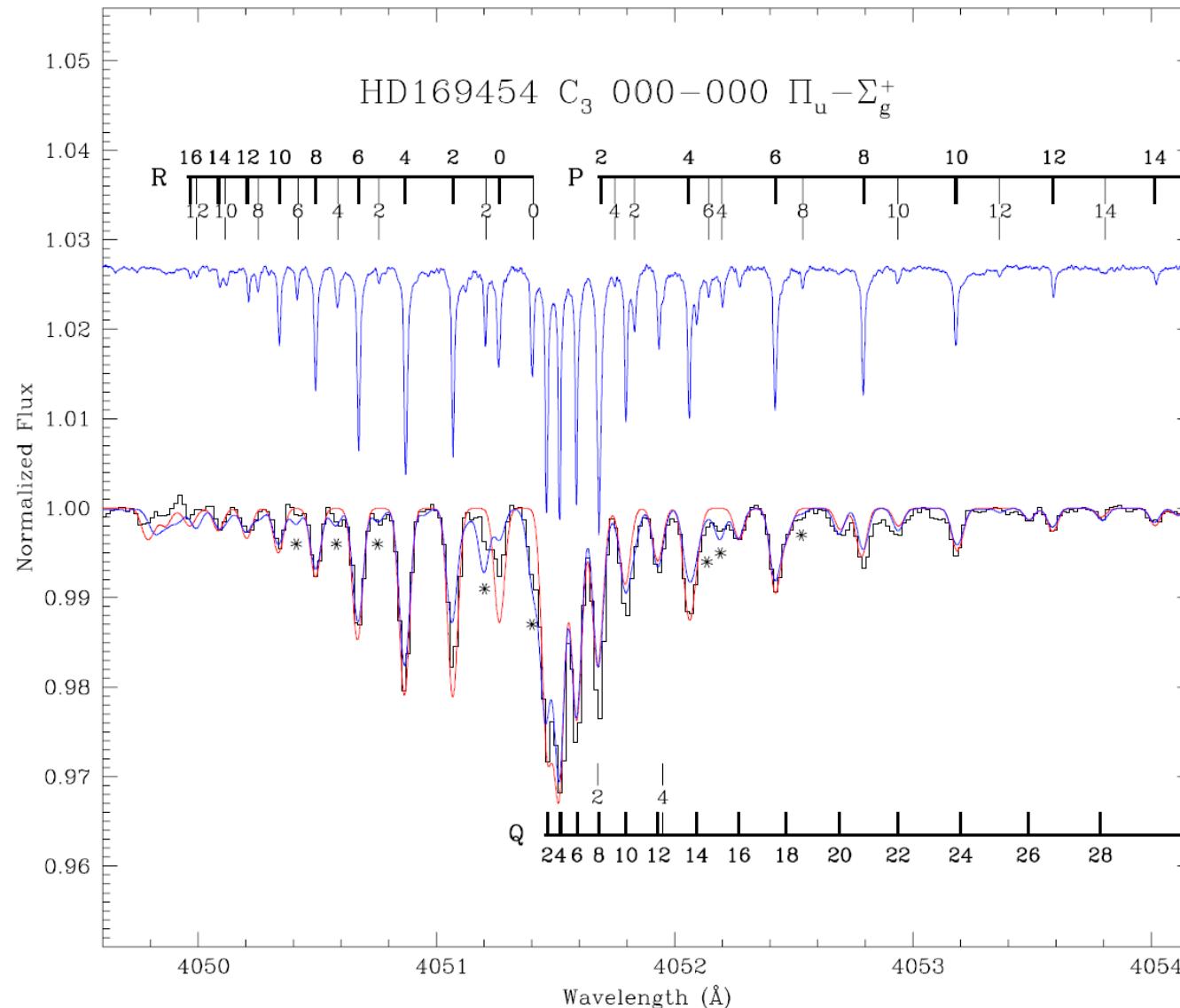


Figure 2. Spectrum of the $C_3 \tilde{\Lambda}^1 \Pi_u - \tilde{X}^1 \Sigma_g^+$ 000–000 band. Upper spectrum: laboratory measurement by cavity ring-down spectroscopy in a planar plasma jet. Lower spectrum: observation in the sightline to HD 169454. Positions of rotational lines are tagged with thick lines. Thin lines tag rotational lines due to perturbing states based on the analysis of Zhang et al. (2005). The astronomical spectrum is overlaid with fitted curves: the red curve represents a fit using only unperturbed lines; for the blue curve perturber lines are included in the fit. Note the remaining deviation for the intensity of the R(0) line in the simulated spectrum (see text).

Table 7. Summary of observed molecular column densities towards HD 169454.

Molecule	N_{col} (10^{12} cm^{-2})	T_{exc} (K)	Source
C_2	65 ± 1	19 ± 2	1
	73 ± 14	15^{+10}_{-5}	2
	70 ± 14		3
	160 ± 29		4
C_3	6.61 ± 0.19	22.4 ± 1.0	5
	2.24 ± 0.66^a	23.4 ± 1.4	4
	4.5 ± 0.3^b		3
CH	39.6 ± 0.3^c		5
	46 ± 8		2
	$36.5^{+12.6}_{-7.8}$		6
CH^+	20.8 ± 0.2^e		5
H_2	$(8 \times 10^{20}^d)$		5

^aBased on a sample of rotational lines with $J \leq 8$.

^bBased on unresolved rotational lines.

^cBased on CH line at 3886.409 Å.

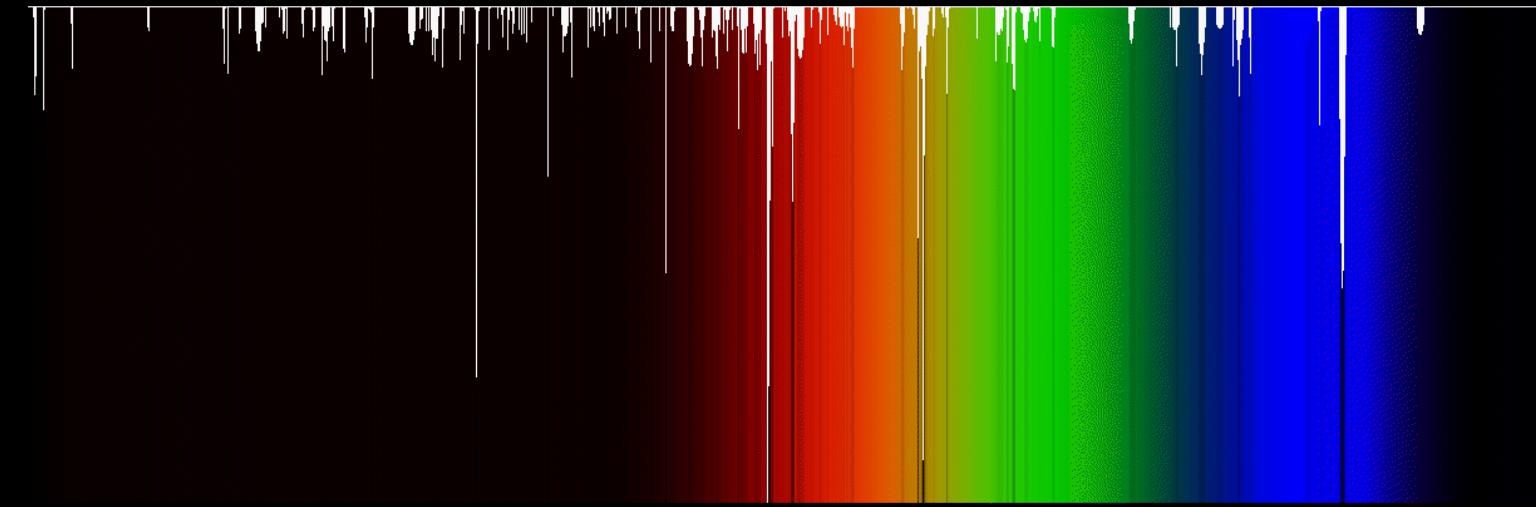
^dBased on correlation of CH w.r.t. H₂ (Weselak et al. (2004); fig. 3. therein).

^eBased on the CH⁺ line at 3957 Å.

Source: (1) Kazmierczak et al. (2010a); (2) Jannuzi et al. (1988); (3) Oka et al. (2003); (4) Ádámkovics et al. (2003); (5) This work; (6) Crawford (1997).

Schmidt et. al,
MNRAS **441**,
1131(2014)

The Diffuse Interstellar Bands



Courtesy: P. Jenniskens, F.-X. Desert

Absorption features seen in the spectra of astronomical objects in the Milky Way and other galaxies. They are caused by the absorption of light by the interstellar medium. About 500 bands have now been seen, in UV, visible, and IR.

The origin of DIBs was unknown and disputed for many years, and the DIBs were long believed to be due to polycyclic aromatic hydrocarbons and other large carbon-bearing molecules.

Schmidt et. al,
MNRAS 441,
1131(2014)

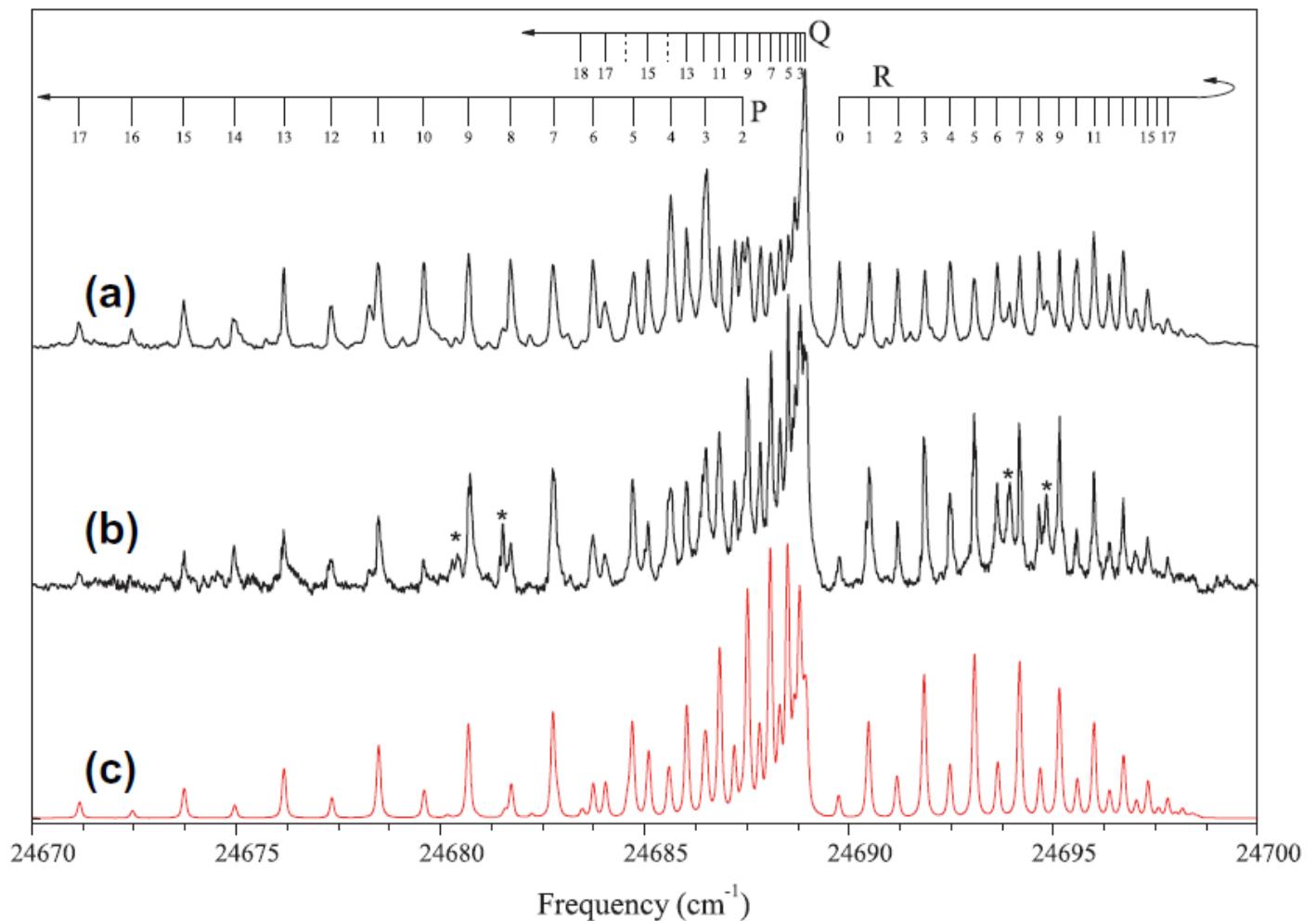


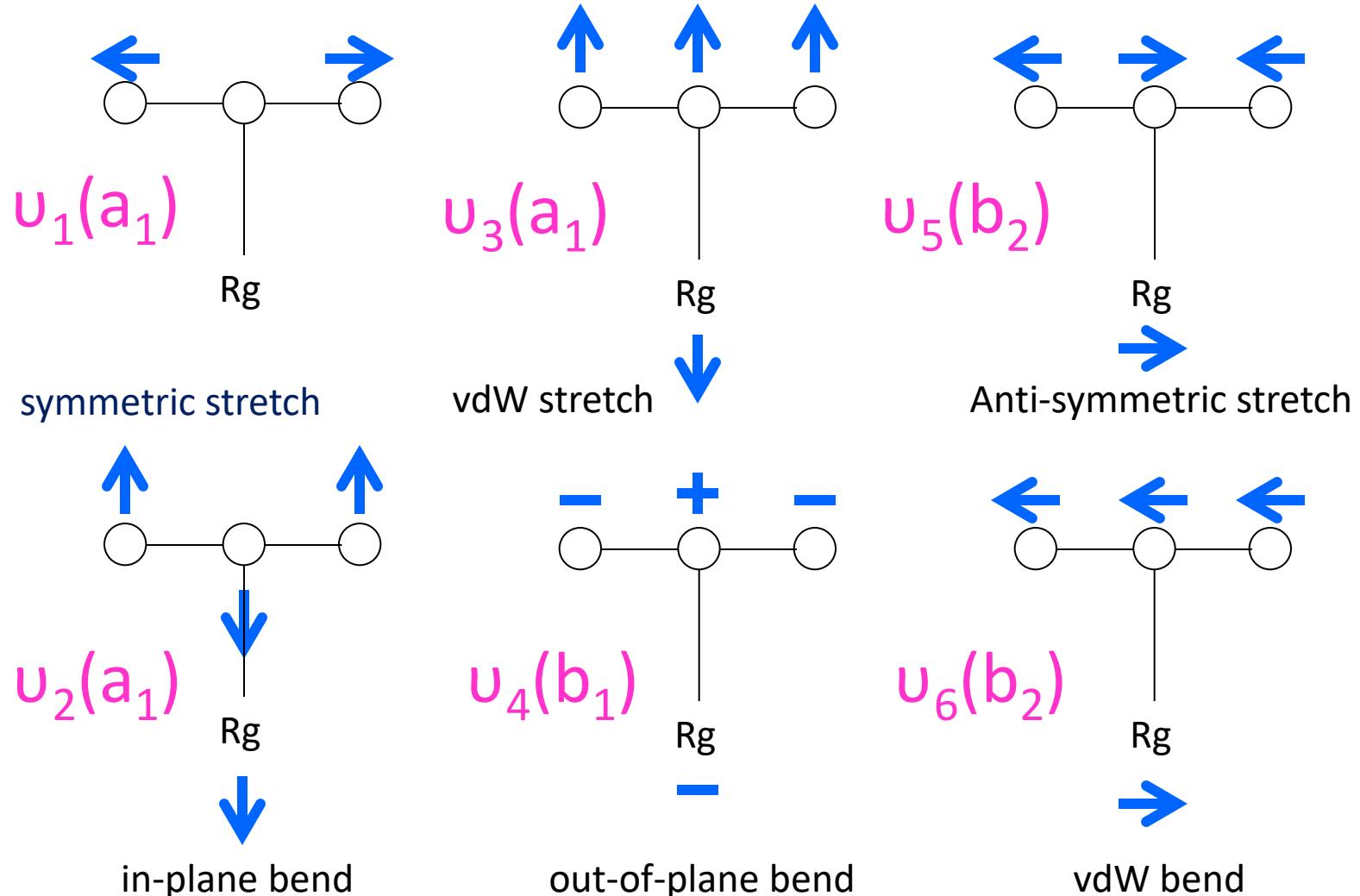
Fig. 2. Rotationally resolved spectra of the $\tilde{\Lambda}^1\Pi_u - \tilde{X}^1\Sigma_g^+$ electronic origin band of $^{13}\text{C}_3$. (a) Saturated spectrum recorded with $\sim 0.2\%$ of $^{13}\text{C}_2\text{H}_2$ in the gas mixture; (b) Spectrum recorded with $\sim 0.03\%$ of $^{13}\text{C}_2\text{H}_2$ in the gas mixture yielding reliable intensities. (c) The red trace shows the simulated spectrum for a $^1\Pi_u - ^1\Sigma_g^+$ transition, with derived constants for $^{13}\text{C}_3$, a Lorentzian width of 0.07 cm^{-1} , a Gaussian width of $\sim 0.1\text{ cm}^{-1}$, and an estimated rotational temperature of $\sim 45\text{ K}$. The lines marked with an asterisk are due to blending transitions of other small species like $^{13}\text{C}_2$ or ^{13}CH , etc. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

No perturbation was observed in their $^{13}\text{C}_3$, 000-000 band (lab), consistent with our assignments. Perhaps next generation telescope would allow us to observe the transition of $^{13}\text{C}_3$.

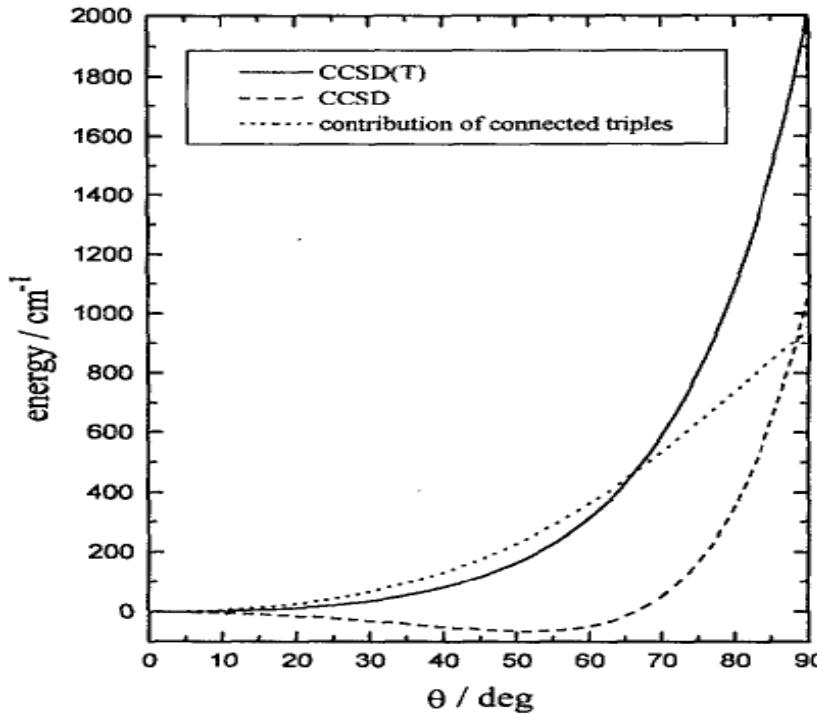
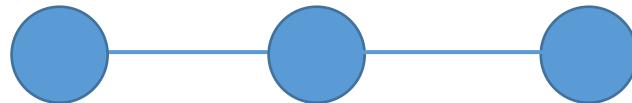
Laser spectroscopic studies of C₃Ar van der Waals complex

1. G. Zhang, B.-G. Lin, S.-M. Wen, and Y.-C. Hsu, J. Chem. Phys. **120**, 3189-3200(2004).
2. A. J. Merer, Y.-C. Hsu, Y.-R. Chen, and Y.-J. Wang, J. Chem. Phys. **143**, 194304(2015).
3. Y.-J. Wang and Y.-C. Hsu, (to be published).

The Normal Modes of C₃-Rg Complex



Unique feature of $\text{C}_3(\tilde{\text{X}})$



Unusually low bending frequency
 63 cm^{-1}
Schmuttenmaer et. al., Science (1990)

Mladenović et.al., J. Chem. Phys.
101, 5891 (1994).

How would the rare gas atom affect the large amplitude motion of C_3 ?

Unique features of $C_3(\tilde{\Lambda})$



Non-equivalent C-C bond lengths was observed in the $\tilde{\Lambda}$ state.

Large Renner Effect was reported in the $\tilde{\Lambda}$ state (Gausset et. al., Astrophys. J. 1965)

Due to the static effect set up by molecular bending motion and π - orbital, the degeneracy of the Π state is lifted.----- **A failure of B-O approximation.**

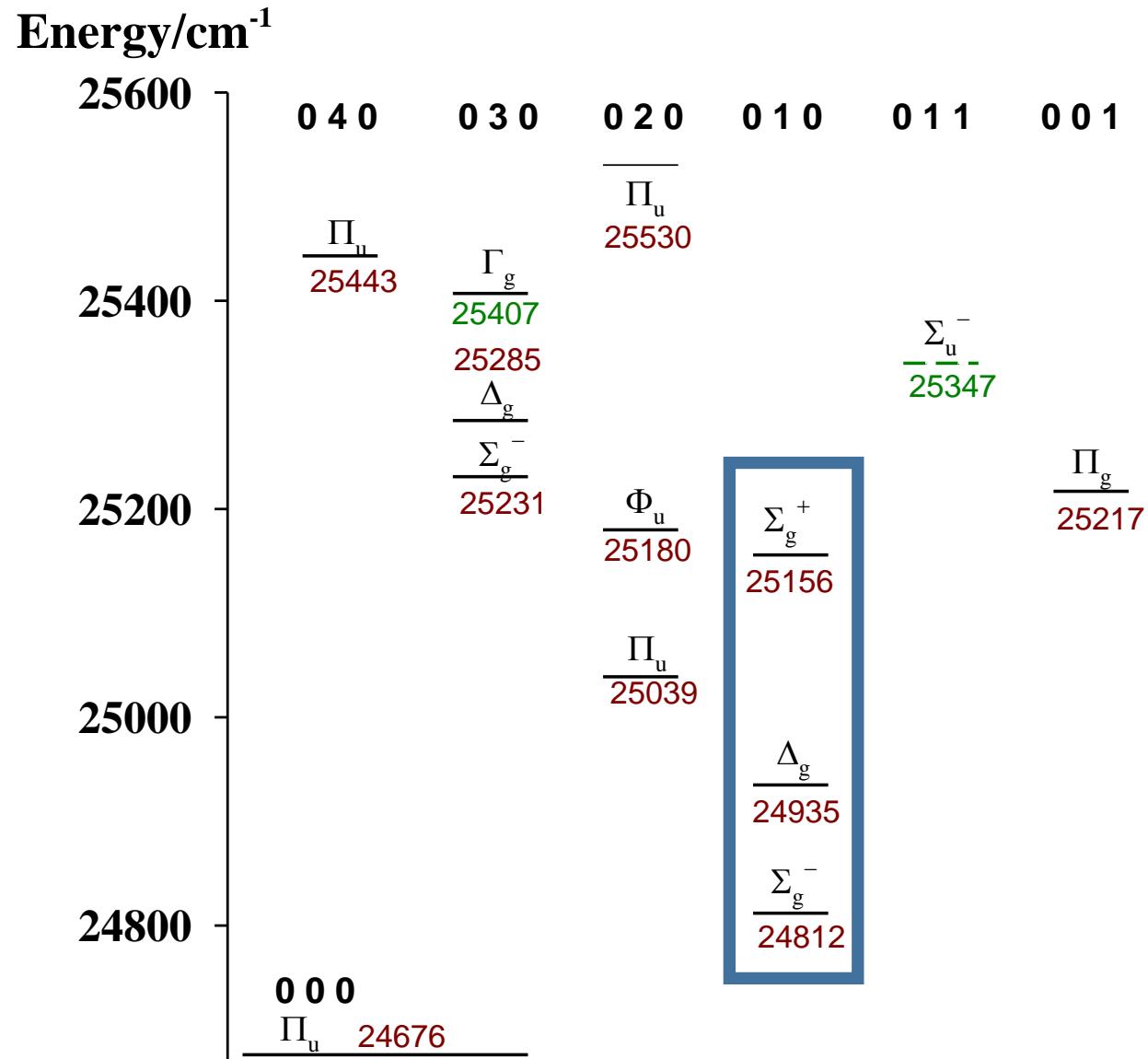
$K = \Lambda + \ell$, *good* quantum number

K : projection of total angular momentum onto the principle axis of C_3

Λ : projection of electronic angular momentum onto the principle axis of C_3

ℓ : vibrational angular momentum

The Renner-Teller Effect of the $\tilde{\Lambda}$ state of C_3

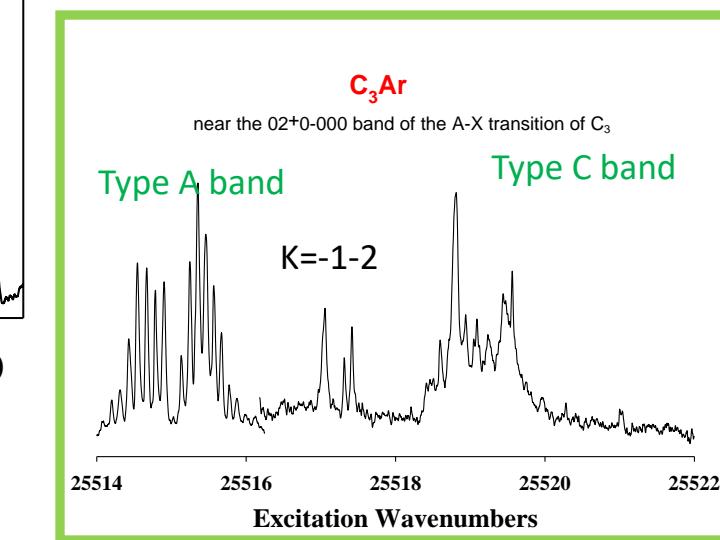
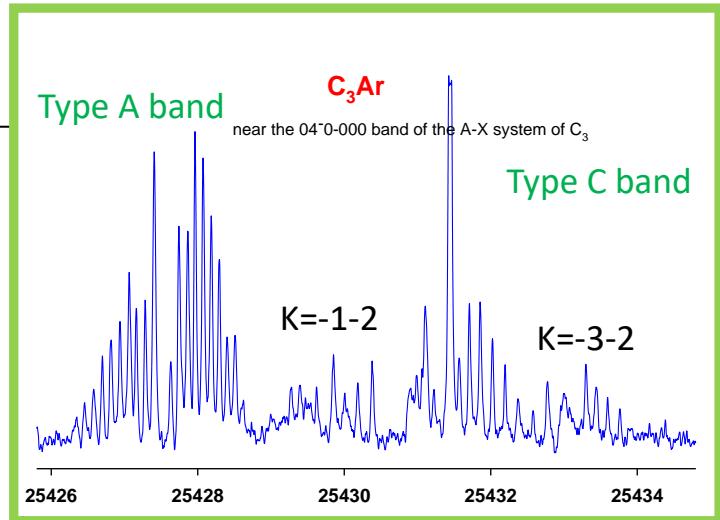
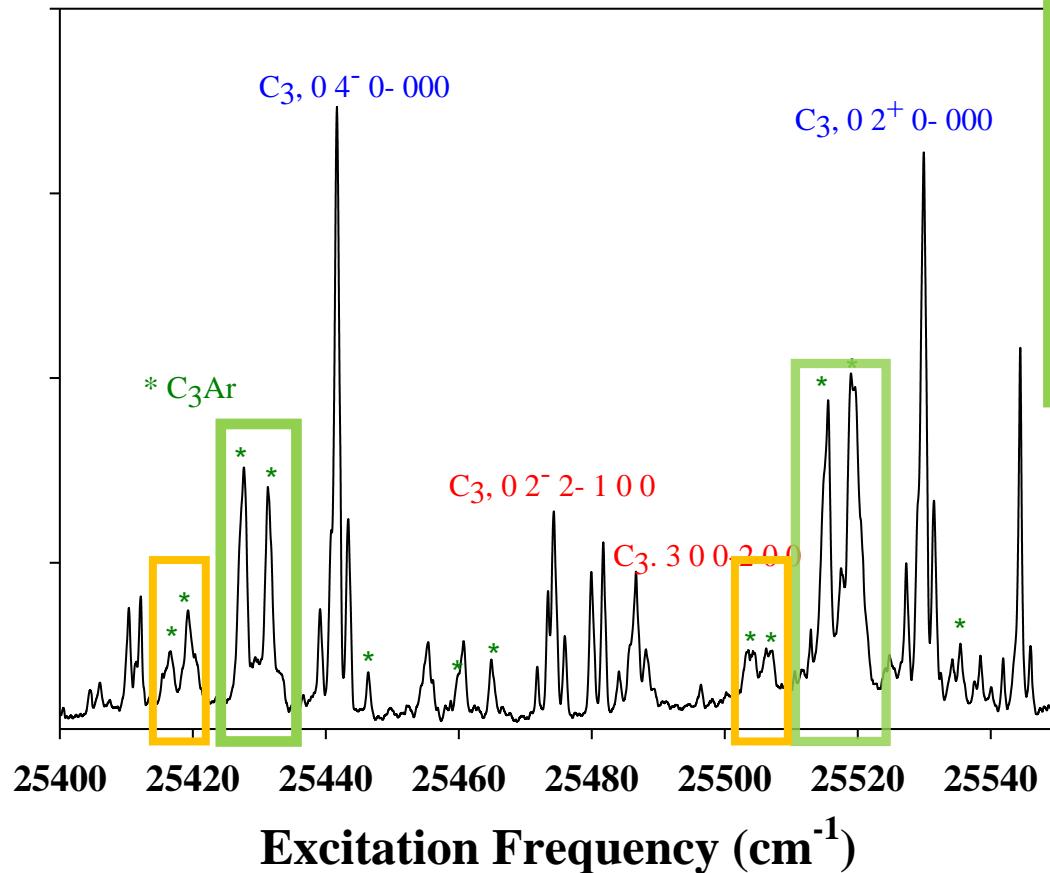


$K = \Lambda + \ell$, *good* quantum number
K: projection of total angular momentum
onto the principle axis of C_3
 Λ : projection of electronic angular
momentum onto the principle axis of C_3
 ℓ : vibrational angular momentum

Chen et. al., J. Mol. Spectrosc. **267**, 169-171 (2011).

LIF Excitation Spectra of C_3Ar and C_3

- *In the region 25410-25535 cm^{-1}*



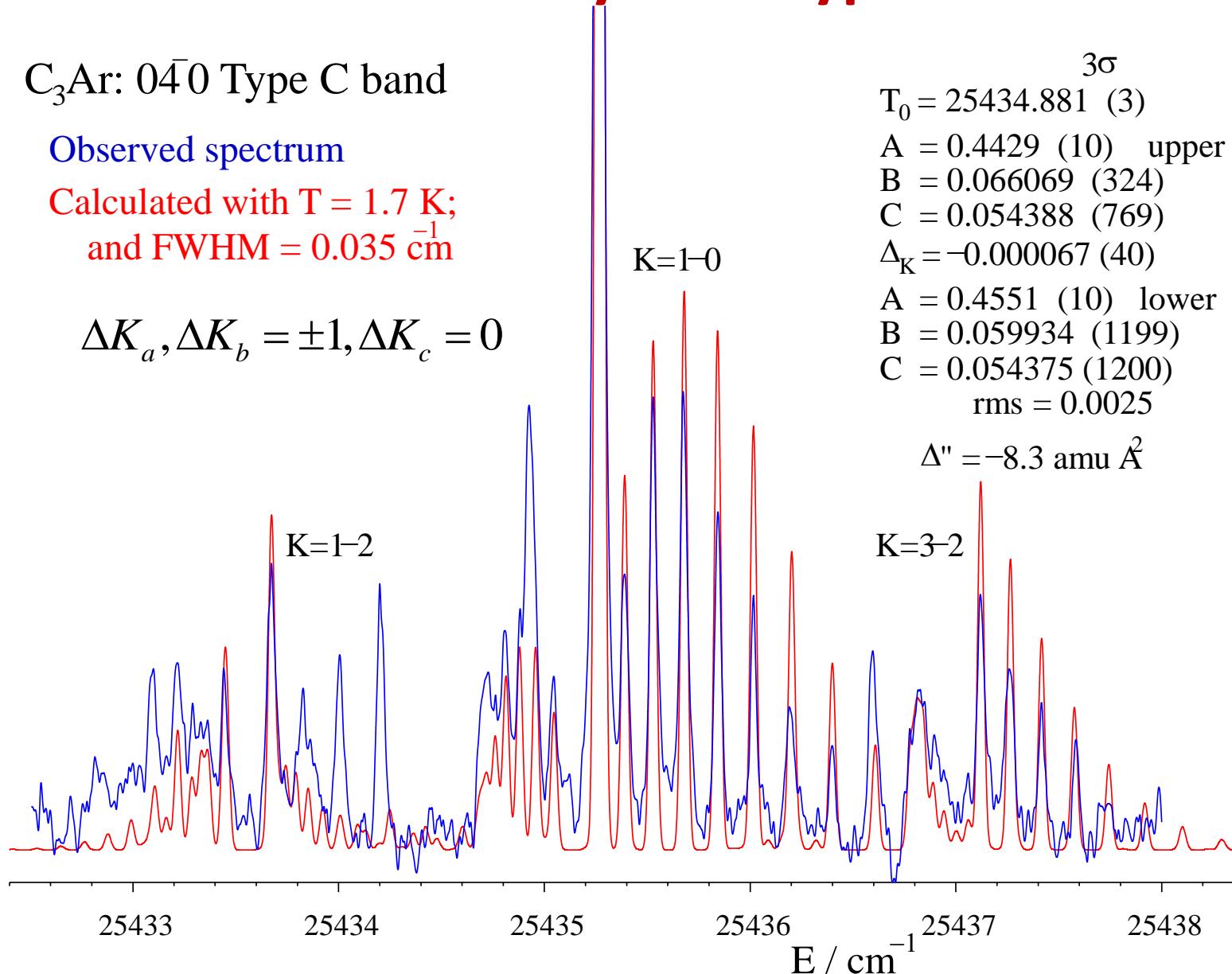
Band Contour Analysis of Type C band

C₃Ar: 04̄0 Type C band

Observed spectrum

Calculated with T = 1.7 K;
and FWHM = 0.035 cm⁻¹

$$\Delta K_a, \Delta K_b = \pm 1, \Delta K_c = 0$$



Band Contour Analysis of Type A band

C₃Ar: 04⁻0 Type A band

Observed spectrum

Calculated, with T = 1.7 K
and FWHM = 0.035 cm⁻¹

3σ

T₀ = 25431.334 (4)

A' = 0.4284 (16)

B' = 0.059156 (2330)

C' = 0.054934 (1805)

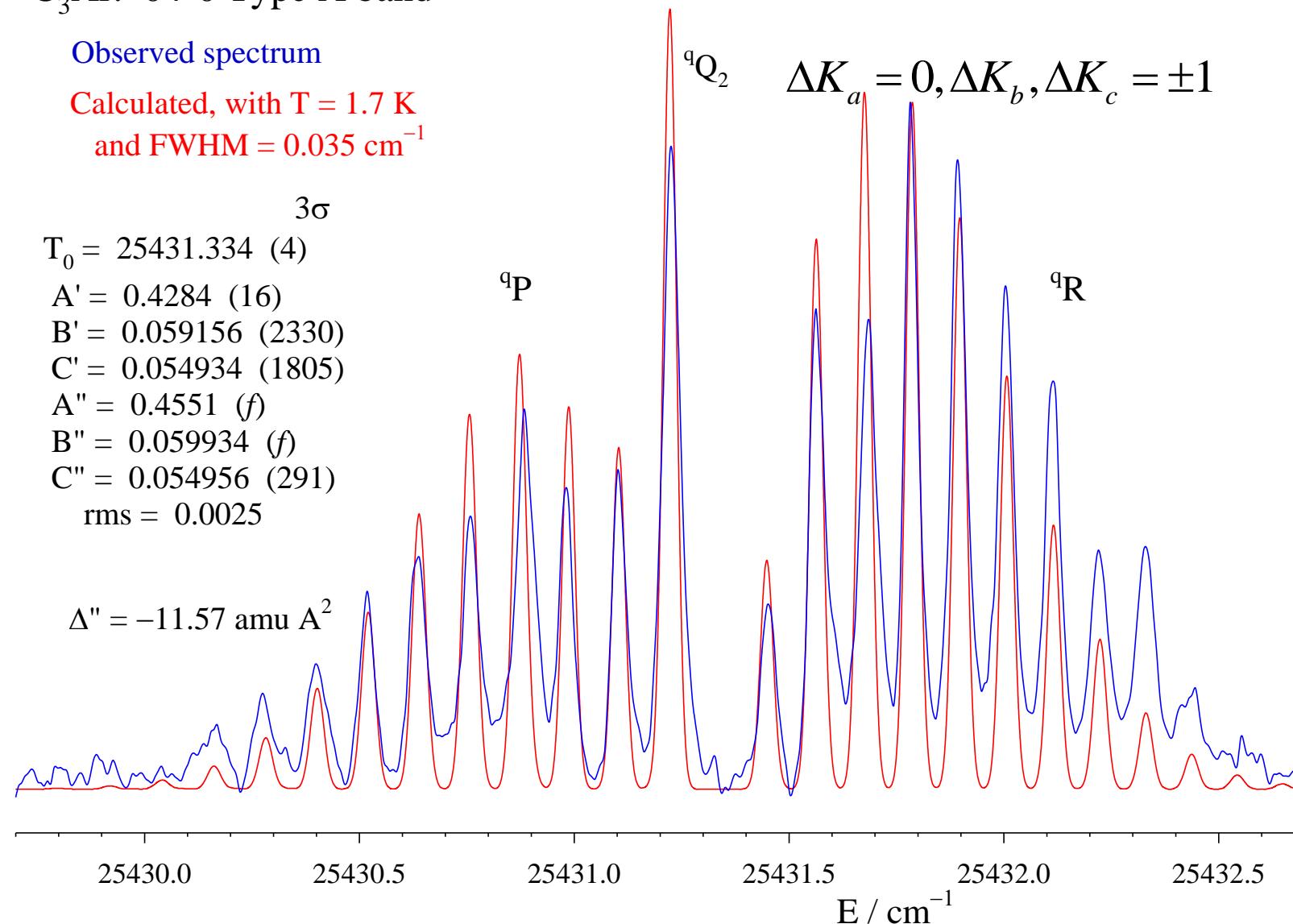
A'' = 0.4551 (f)

B'' = 0.059934 (f)

C'' = 0.054956 (291)

rms = 0.0025

Δ'' = -11.57 amu Å²



Summary of the ROTATIONAL information of the v=0 of C₃Ar(\tilde{X})

1. K''=0 and 2 levels were observed in the LIF spectra.

The ¹²C nucleus is a boson, therefore the Ar atom must lie in the C_{2v} axis of C₃Ar, T-shaped.

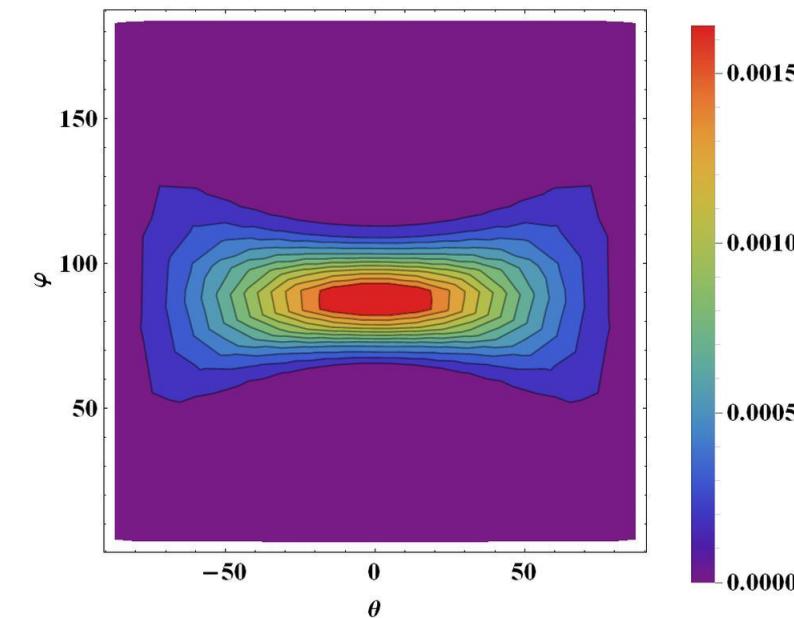
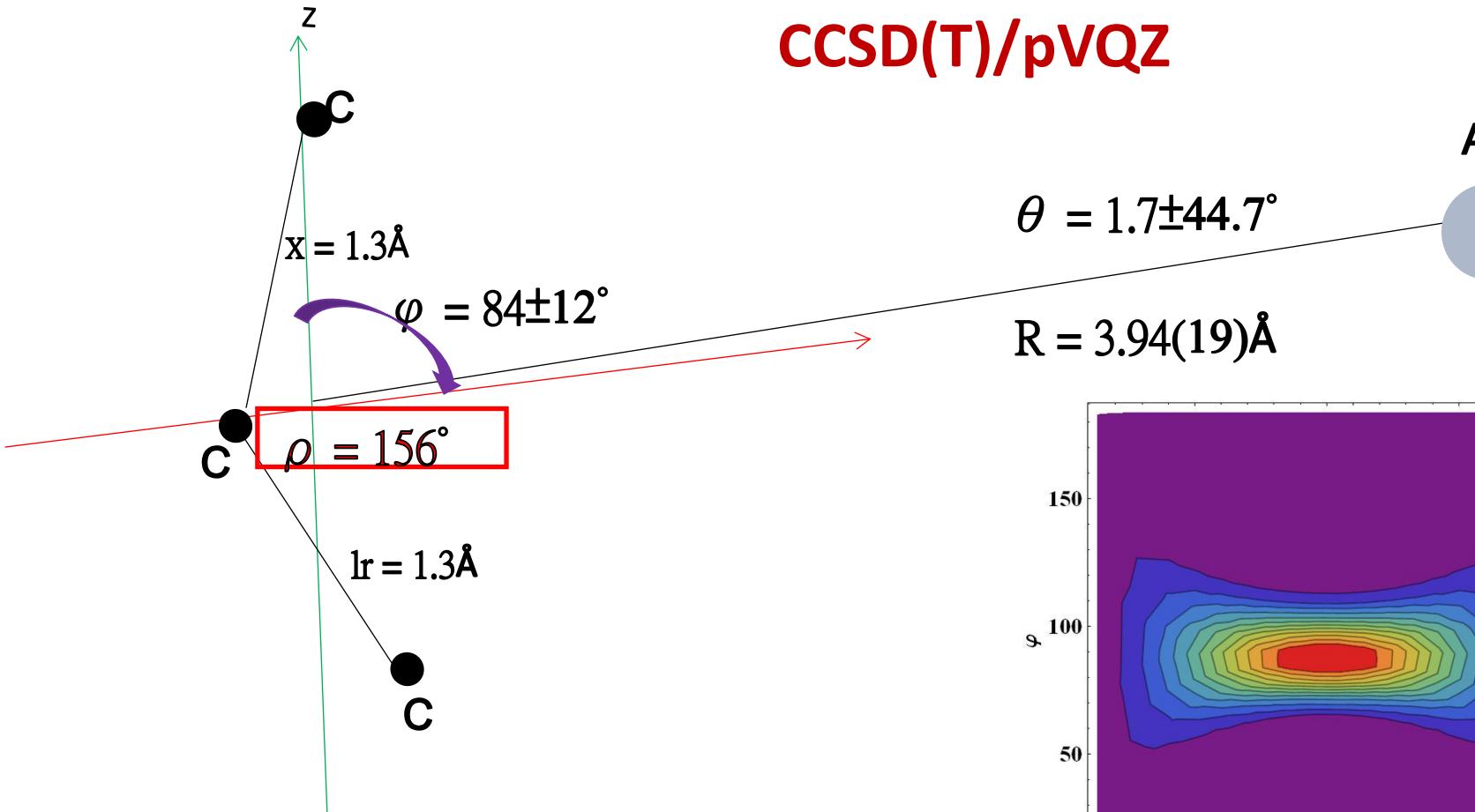
2. From rotational analysis, the rotational constant A is greater than the rotational constant B of C₃.

Experimentally obtained rotational constants are consistent with those calculated from the predicted wavefunctions by MCTDH within 95%.

The rotational constants B and C give the vdW bond length as 3.81Å .

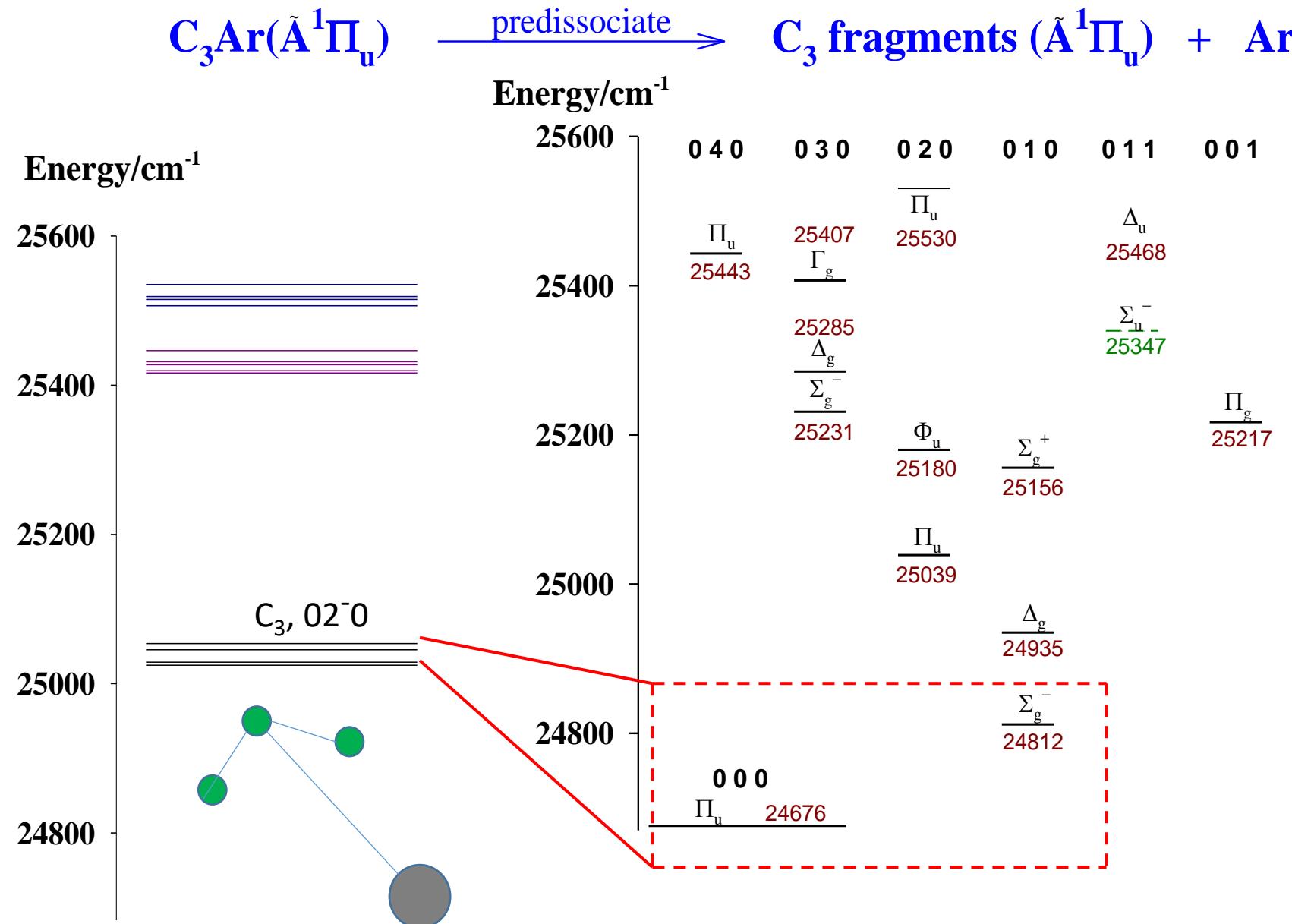
3. The electronic degeneracy (¹Π_u) of the upper state is lifted in the lower symmetry of the complex. Each Π_u vibronic upper level splits into an A₁ and B₁ pair.

The geometry and probability of the v=0 of C₃Ar(\tilde{X})
 from the MCDTH calculation using an ab initio potential at the level of
 CCSD(T)/pVQZ

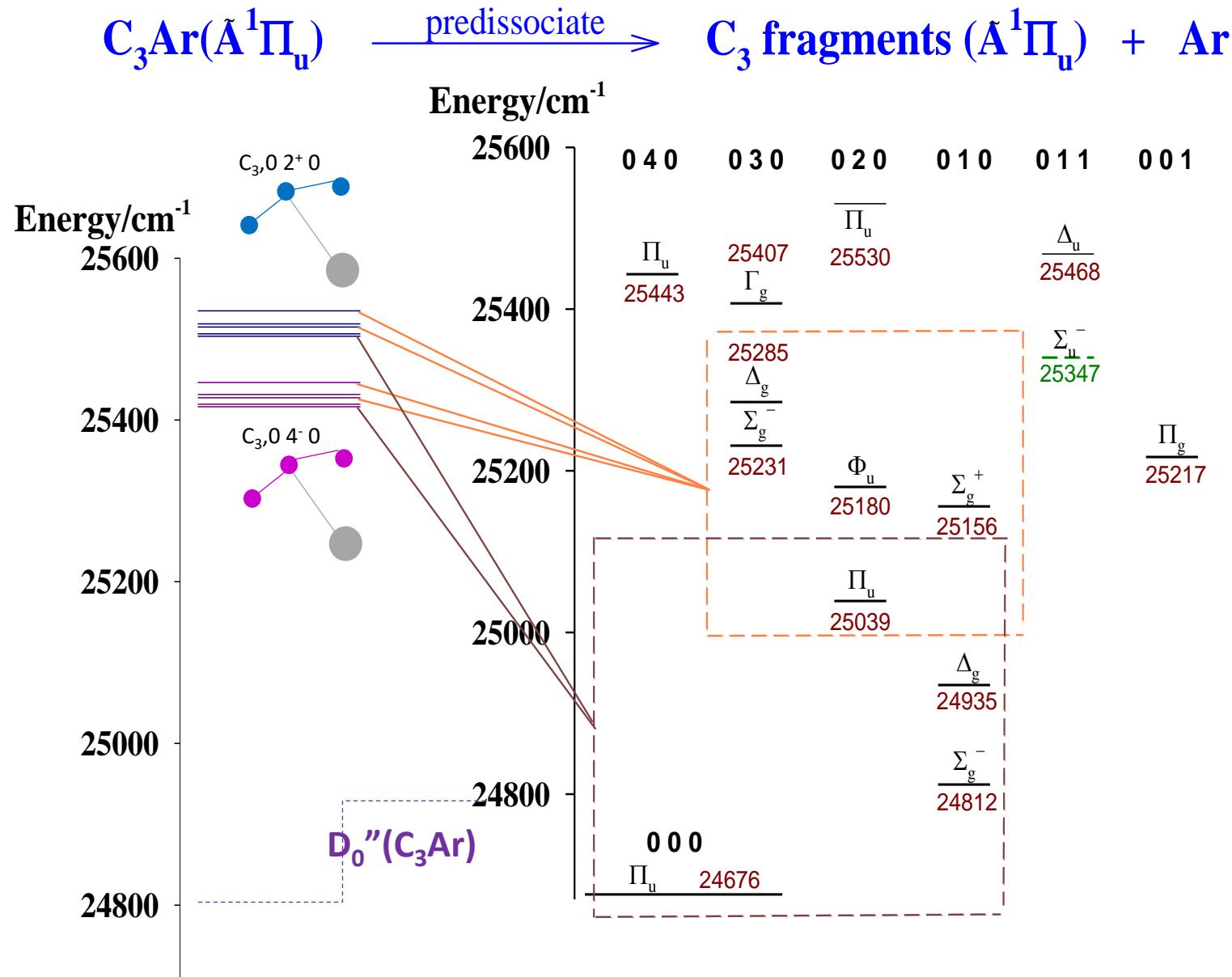


The contour plot of probability distribution was obtained at
 R 3.814 Å , r₁=1.3002 Å , r₂=1.3071 Å , ∠ C-C- C=156.9°

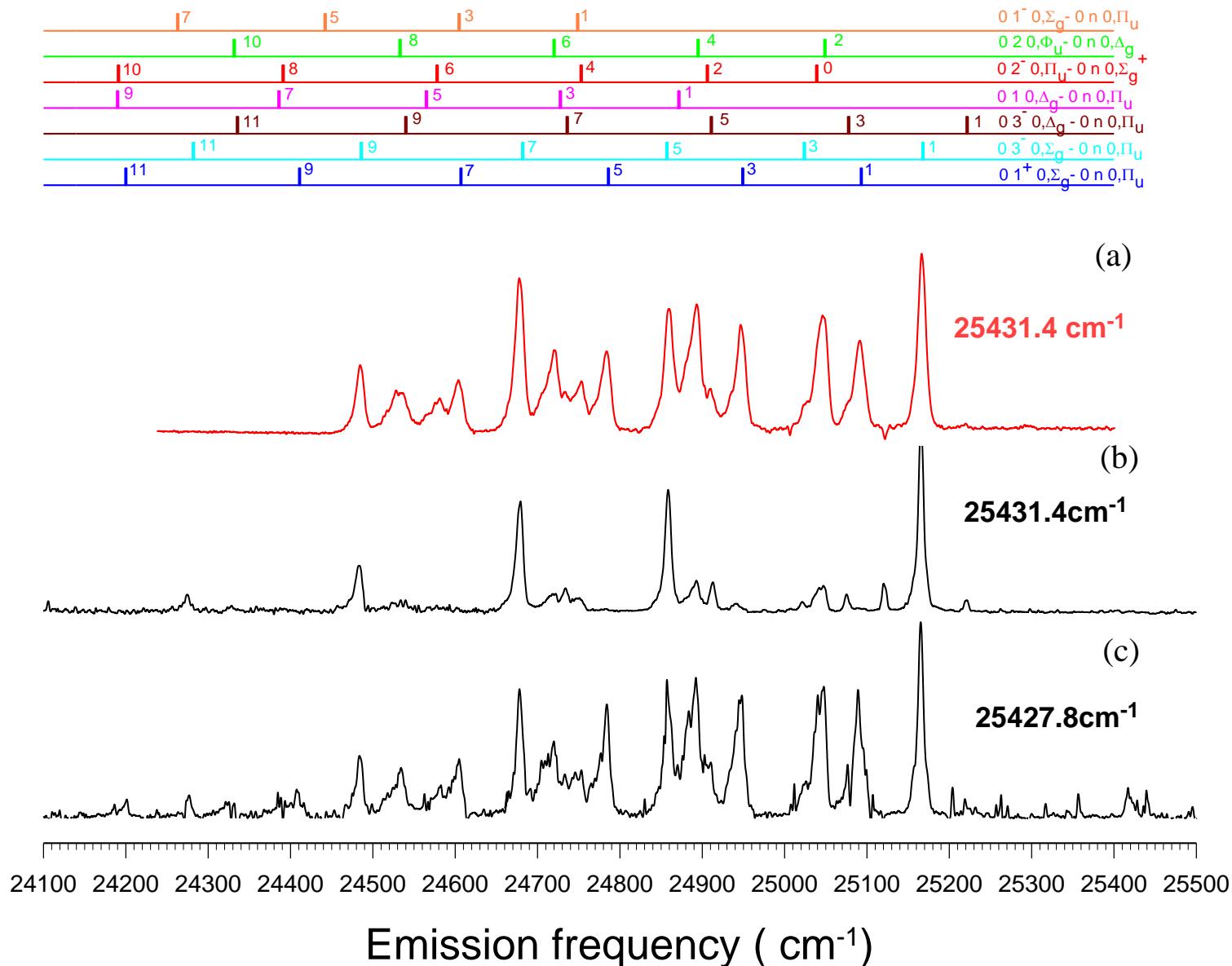
• Predissociation pathways of C_3Ar near the 25039 cm^{-1} region



• *Predisociation pathways of C_3Ar near the 25416-25520 cm^{-1} region*

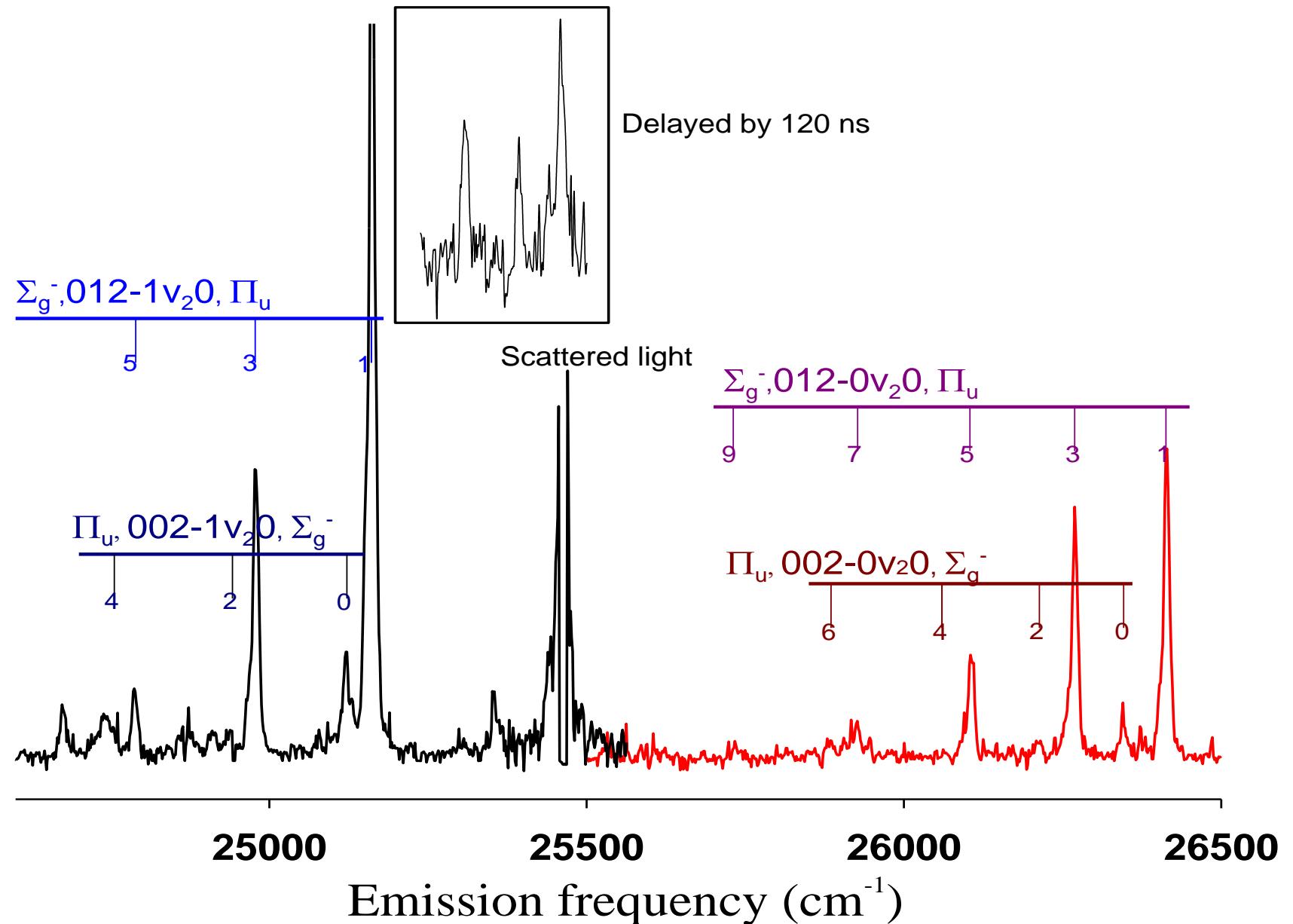


Relaxation observed in the emission spectra of C₃Ar from the 25431 cm⁻¹ band



C_3Ar relaxation observed from the 26689 cm^{-1} level near the $C_3, 02^-2-100$ band

- $E(A, 012) = 26476\text{ cm}^{-1}$.
- Although 09^-0 and 01^-2 levels are nearly degenerate. Only $\Delta v_2=1-2$ transfer are prominently seen.



Estimated energy disposals of the predissociation products

Green bands

R 30 cm^{-1}

T $\sim 120 \text{ cm}^{-1}$

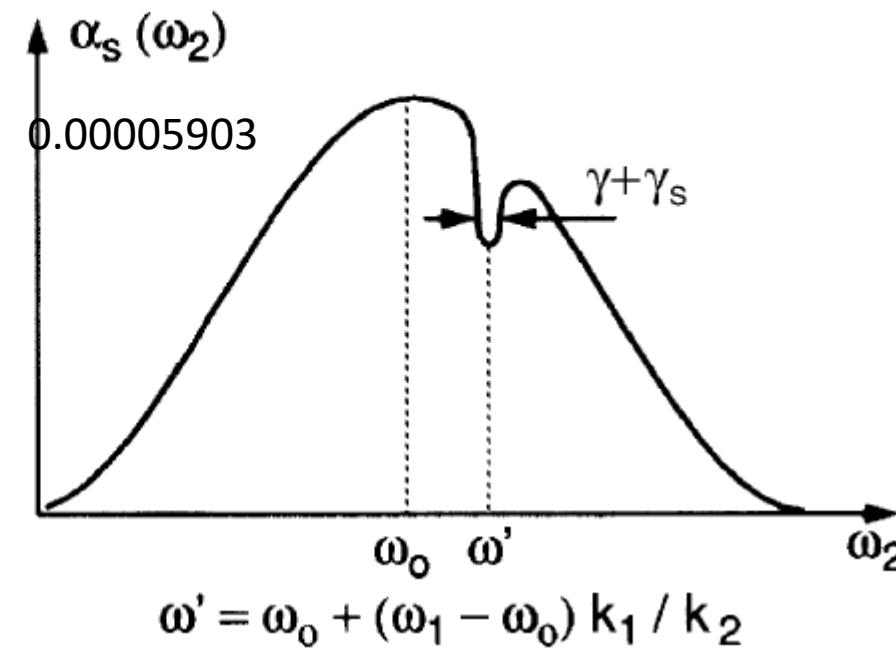
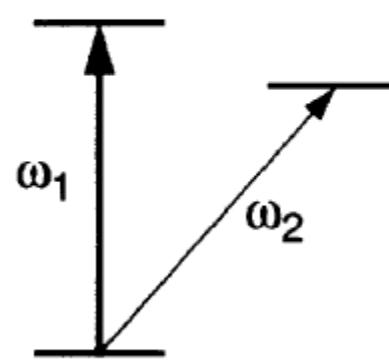
V $\sim 540 \text{ cm}^{-1}$

Orange bands

R+T $\sim 420 \text{ cm}^{-1}$

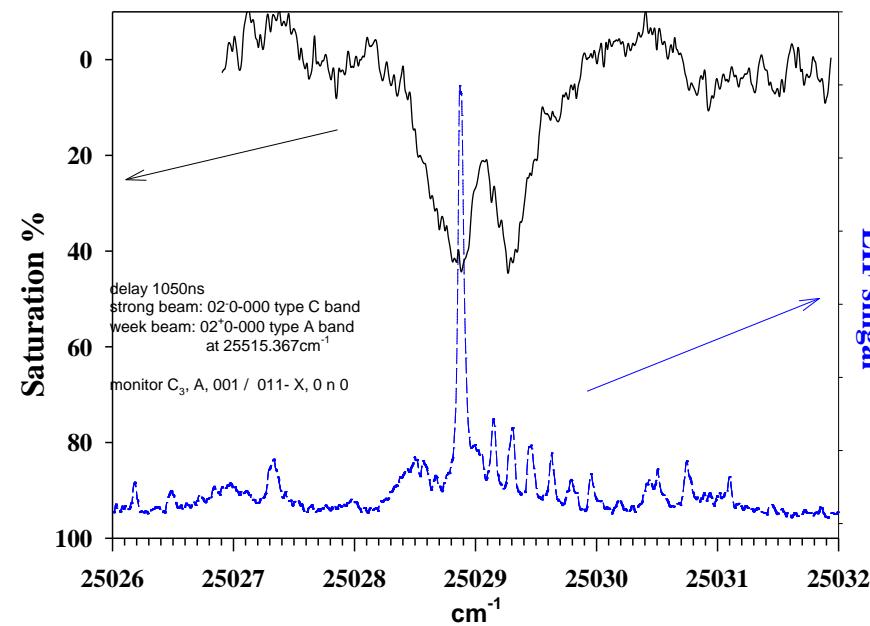
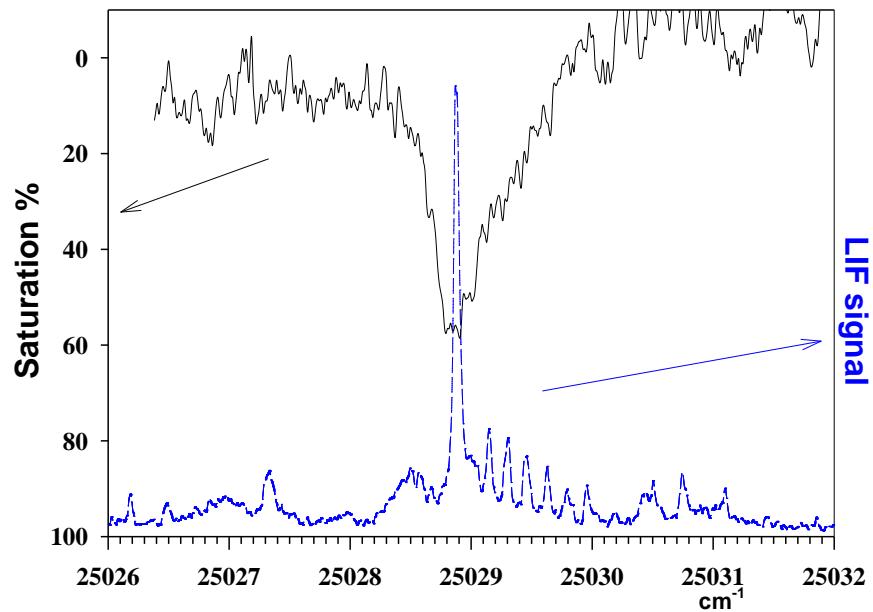
V $\sim 260 \text{ cm}^{-1}$

Hole burning experiment



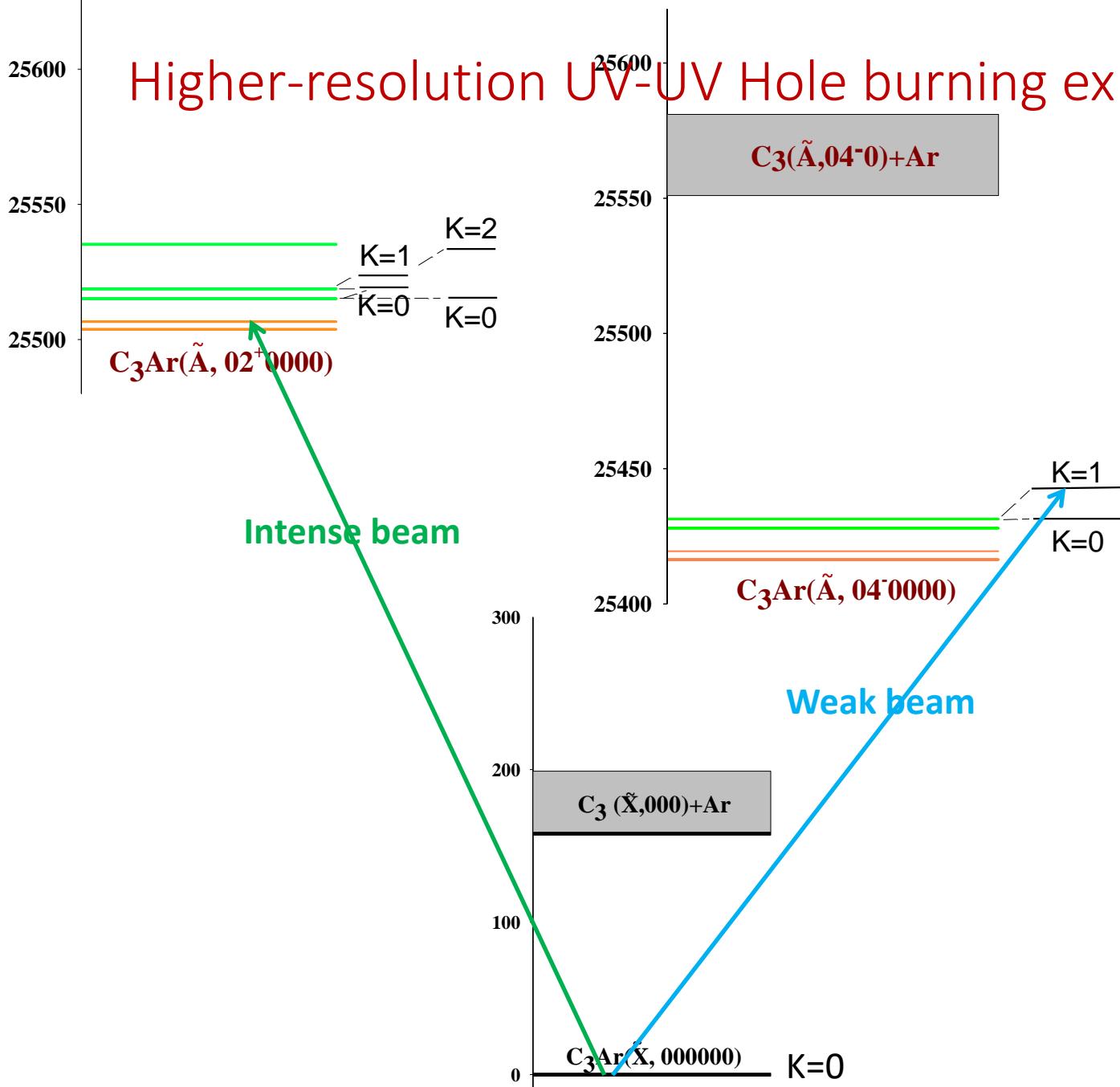
UV-UV Hole burning experiment of the *rotationally resolvable* bands

Scanning the type C band of $\tilde{\Lambda},002^- 000-\tilde{X},000000$ of C_3Ar by monitoring the LIF of the type C/A band of $\tilde{\Lambda},002^+200-\tilde{X},000000$.

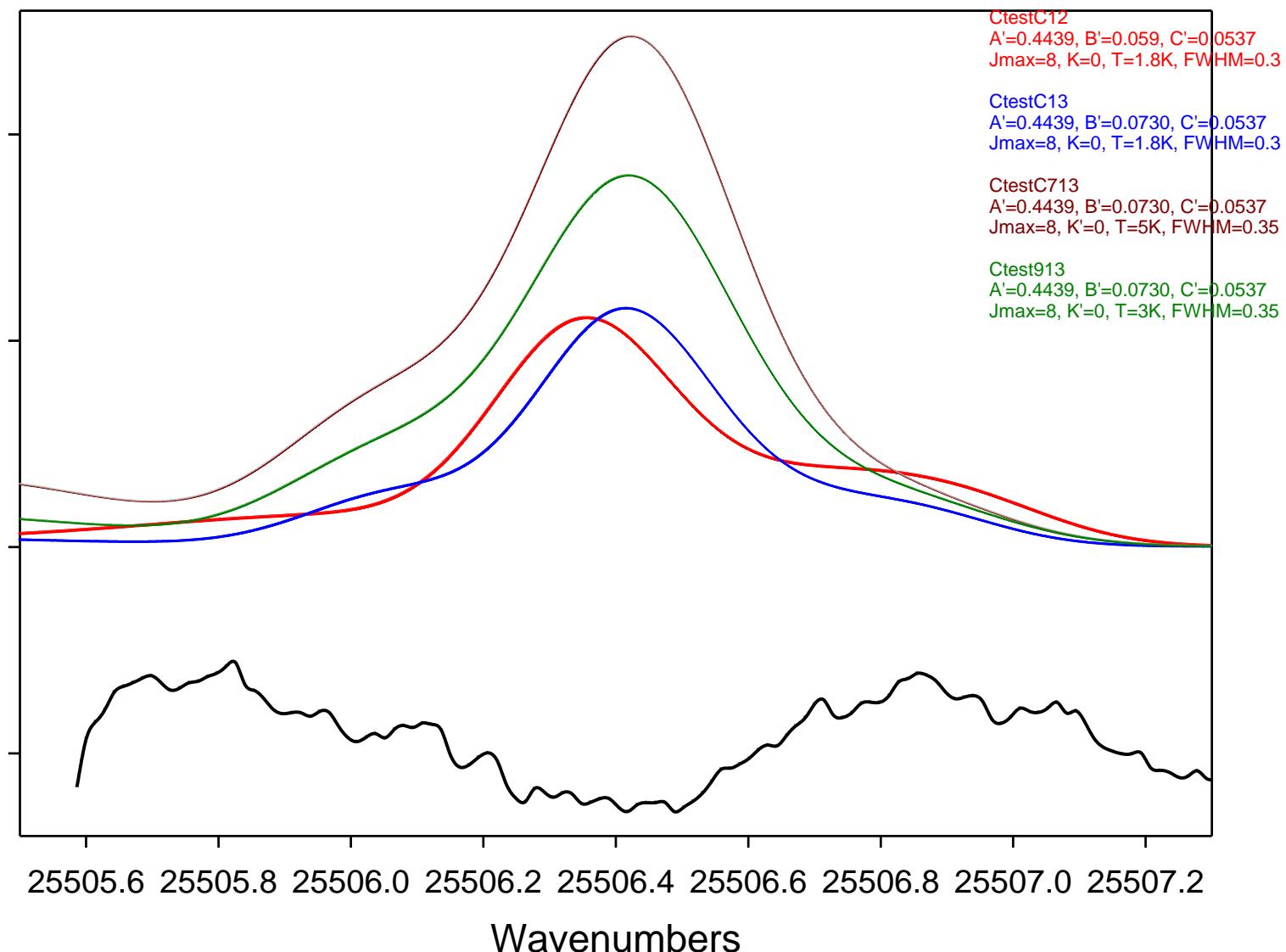


→ Six rotationally resolvable bands are all excited from the lowest level
- - the $v=0$ of $C_3Ar(\tilde{X})$.

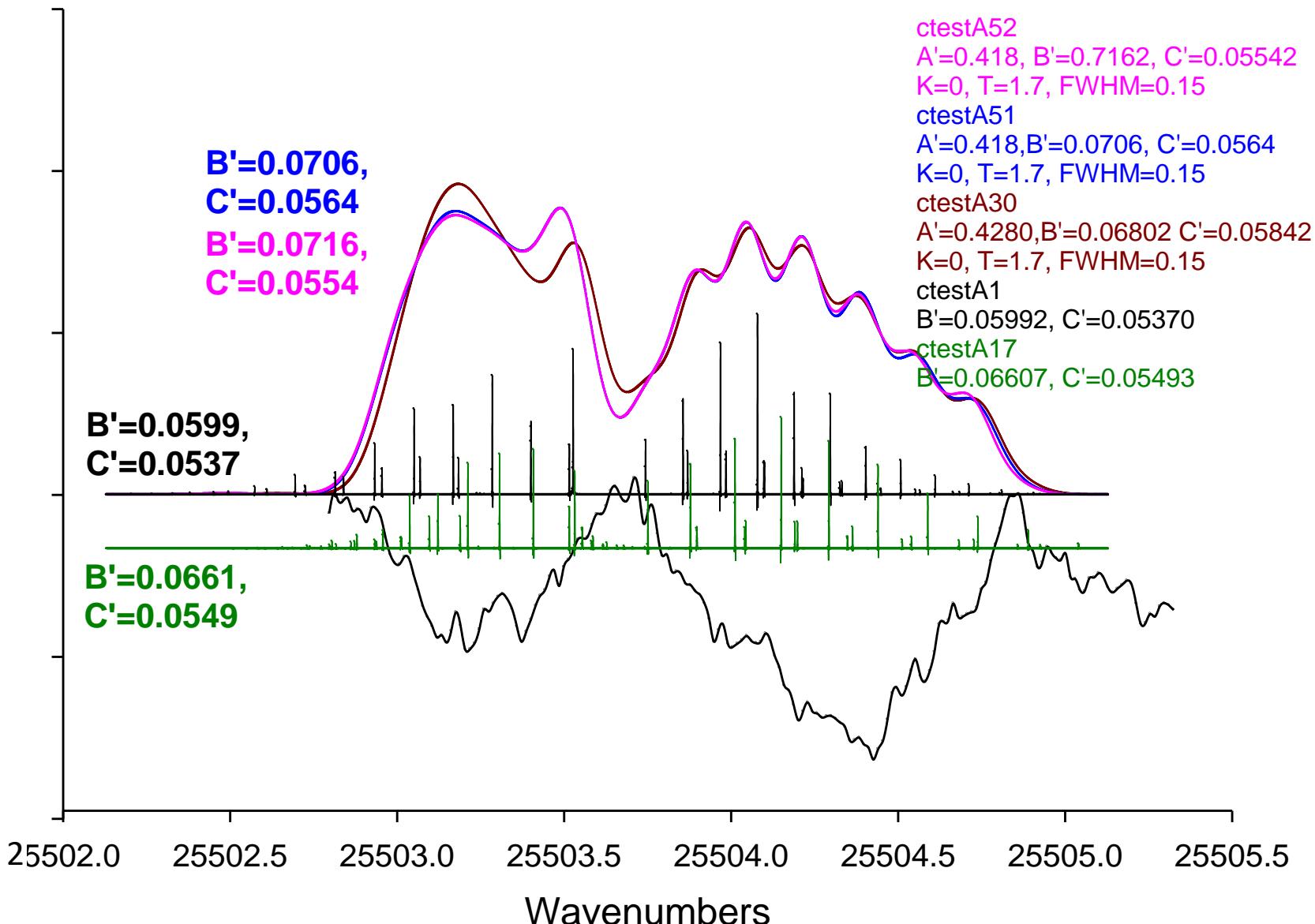
Higher-resolution UV-UV Hole burning experiment

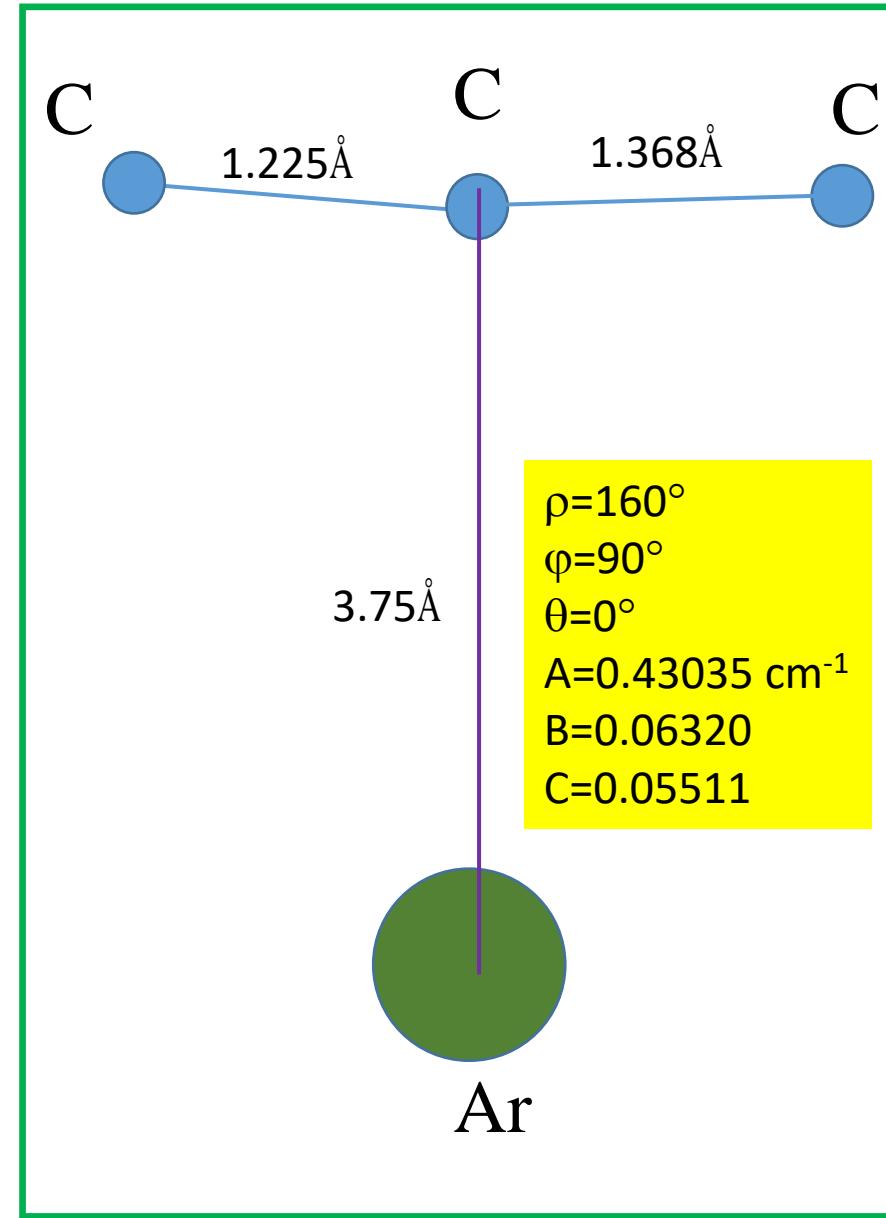
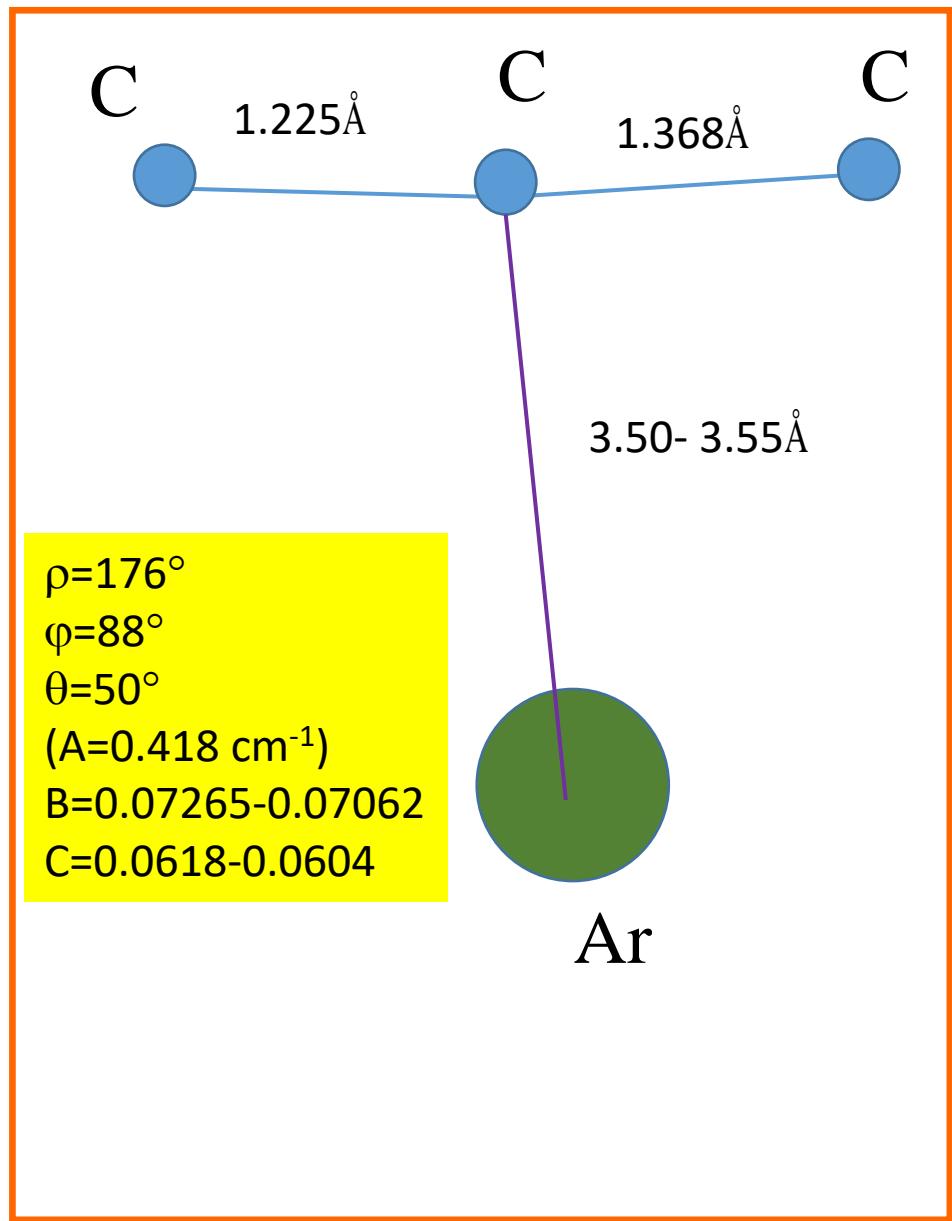


Hole burning spectrum of the 25506 cm^{-1} band



Hole burning spectrum of the 25504 cm^{-1} band





Prospective

- Studies of the C₃Ar levels which have no ν_2' excitations.
- Stimulated emission pumping spectrum of the 25431 cm⁻¹ band to obtain the internal energy of one of the predissociation products, C₃.
- Stimulated emission pumping spectrum of the 26689 cm⁻¹ band to obtain the ground state level structure of the C₃Ar complex.

Acknowledgement

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MOST

